

Estuaries in a changing world

May 18 – 22, 2008

Xiamen, China

2008年5月18 – 22日

中国 厦门

厦门欢迎您！

Xiamen Welcomes You!

High-precision frequency measurements: indispensable tools at the core of molecular-level analysis of complex systems

N. Hertkorn, et al.

HelmholtzZentrum Muenchen
German Research Center for Environmental Health
Institute of Ecological Chemistry,
85758 Neuherberg, Germany

HelmholtzZentrum münchen

Deutsches Forschungszentrum für Gesundheit und Umwelt



outline

this presentation is a **conceptual overview** about **novel high-resolution analytical tools** which provide the capacity to entirely change our **perception of complexity** in any natural system

*detailed **application** of these tools to the molecular-level structural characterization of **natural organic matter (NOM)** will be demonstrated in the **poster session***

outline

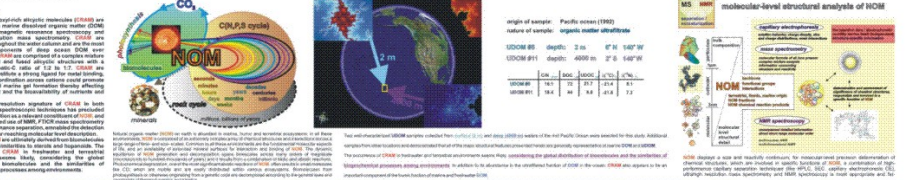
this presentation is a
 about **novel high-res**
 which provide the
 change our **perce**
 in any nat

detailed **application**
 molecular-level stru
 of **natural organic**
 demonstrated in

An integrated NMR and FTICR mass spectroscopic study to characterize a new and major refractory component of (marine) natural organic matter (NOM) at the molecular level, CRAM: carboxyl-rich alicyclic molecules
 N. Hertkorn^a, R. Benner^b, M. Witt^c, M. Frommberger^a, Ph. Schmitt-Koppin^a, K. Kaiser^b, A. Ketrup^a & I. J. Hedges^c
^aGerman Research Center for Environmental Health, Institute of Ecological Chemistry, Ingolstaedter Landstrasse 1, D-85764 Neuherberg, Germany;
^bDepartment of Biological Sciences and Marine Science Program, University of South Carolina, Columbia, SC 29208, USA; ^c Bruker Daltonics, Fahrtenstrasse 4, D-28359 Bremen, Germany; ^dSchool of Oceanography, Box 355351, University of Washington, Seattle, WA 98195-5351, USA

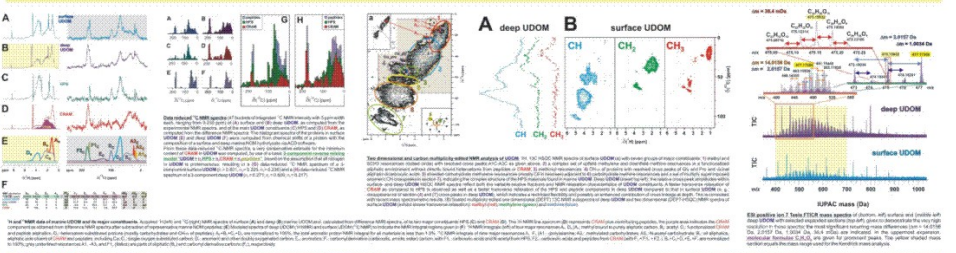
Introduction

ABSTRACT
 Refractory carboxylic aliphatic molecules (CRAM) are identified in marine dissolved organic matter (DOM) using high-resolution mass spectrometry (HRMS) and identified through the combination of the most abundant CRAMs in marine DOM were characterized by 13C and 15N NMR spectroscopy. The molecular weight and structure of CRAMs were determined by comparing the molecular weight and isotopic composition of CRAMs with that of known natural organic matter (NOM) and marine gelatinous biomass. Affinity of CRAMs for various components of DOM was determined by comparing the molecular weight and isotopic composition of CRAMs with that of known natural organic matter (NOM) and marine gelatinous biomass.



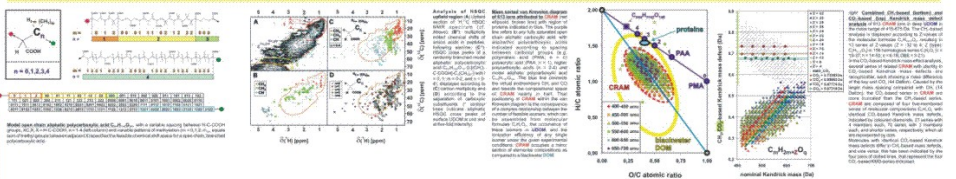
level 1

NMR and MS establish the existence, amount and structural detail of CRAM, which itself is a very complex mixture of carboxylic acids



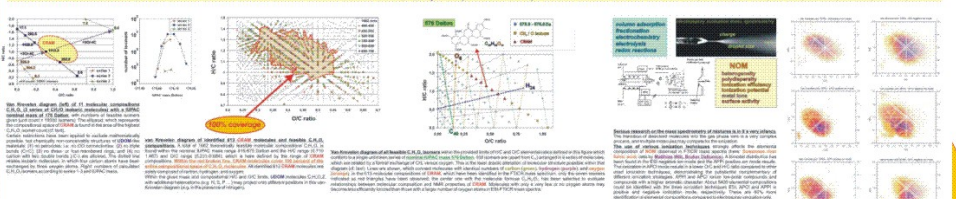
level 2

molecular level detail: CRAM is comprised of alicyclic polycarboxylic acids



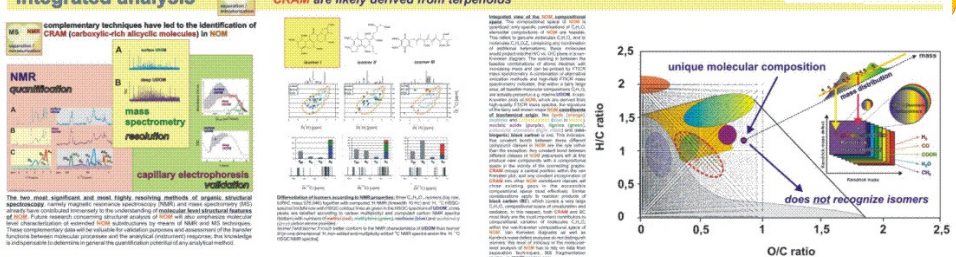
level 3

CRAM cover a substantial fraction of the feasible C,H,O, compositional space, distinct from established biomolecules



integrated analysis

CRAM are likely derived from terpenoids

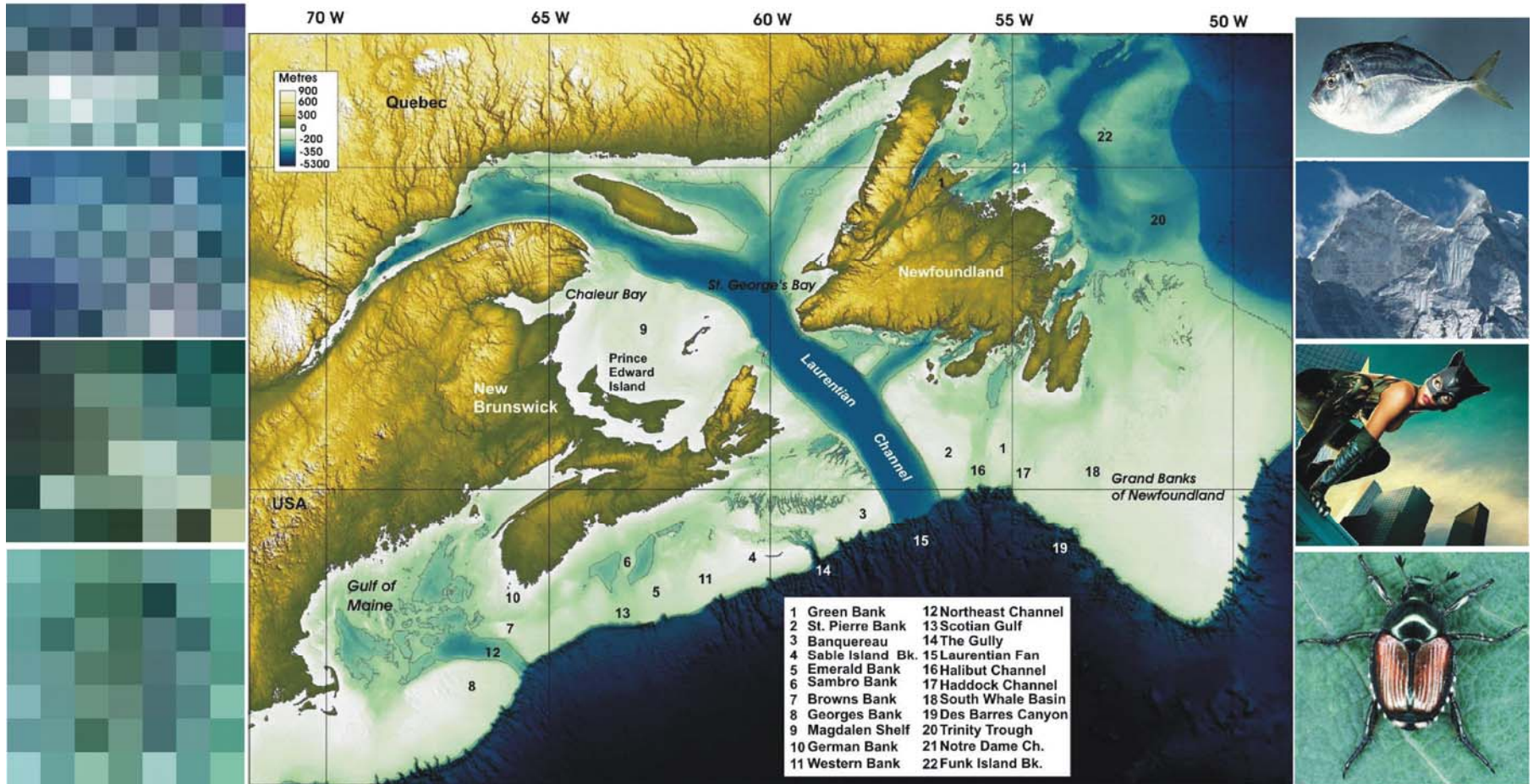


ten-year-vision:

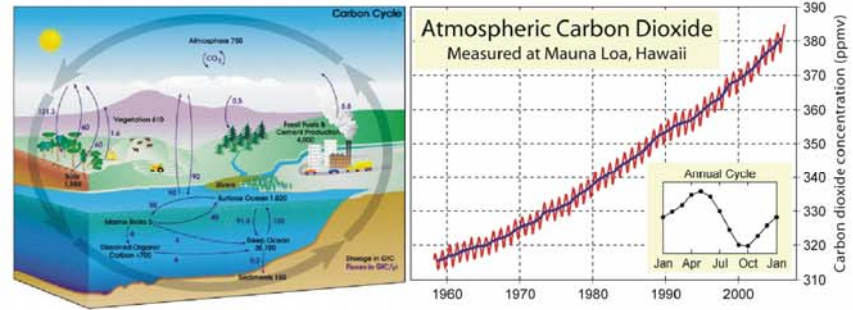
authentic molecular representation of complex natural systems

2005

2015



my "outside view"



CO₂, N₂O, trace gases, FCKW, metal ions



detailed, rich data and models are available

terrestrial, freshwater, marine and atmospheric natural organic matter (NOM)

A diagram showing the "Water Column" and "Water-Sediment-Interface". In the water column, "Algae" are shown with "Particle Flux" moving down. Fluxes at the interface include NO₃⁻, Si, O₂, SO₄²⁻, Fe²⁺, Mn²⁺, NH₄⁺, Si(OH)₄, CH₄, and Cl⁻. The x-axis is "Concentration (relative to bottom water concentration)" from 0% to 100%.
A map of the United States with a color scale from yellow (low) to dark brown (high) representing NOM distribution.
A photograph of a lush green forest with many trees and ferns on the ground.

rather pictorial descriptions of the BIG unknown mystery

spatially resolved target analysis

↔ **models**

*determination of atmospheric trace gases
(e.g., CO₂, N₂O, CH₄)*

molecularly resolved non-target analysis

inventory



models



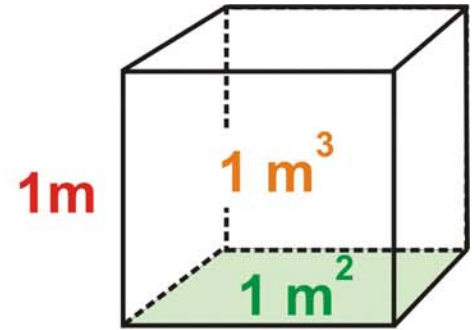
*assessing **molecular composition** and
chemical structures of atmospheric, terrestrial, limnic
and marine organic matter*

spatially resolved target analysis

earth surface: $510.000.000 \text{ km}^2$

dividing the surface in units of $1 \times 1 \text{ m}^2$ and expanding this area into cubes with 1 m^3 size and reaching for 15 km from the surface

$$7.65 * 10^{19} \text{ cubes}$$



determination of concentration in relative units of 10^{-4}

$$7.65 * 10^{19+4} = 7.65 * 10^{23}$$

*options to depict
dissimilarity*

molecularly-resolved non-target analysis

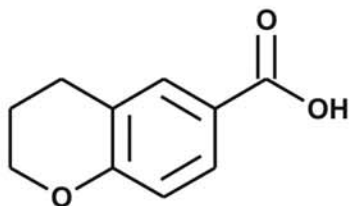
molecular diversity:

number of feasible isomeric molecules in one cubic meter

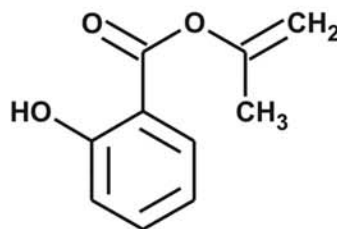
a typical molecule with nominal mass 178 Dalton $C_{10}H_{10}O_3$ (six DBE):

count of chemically relevant isomers: $1.1133 * 10^7$

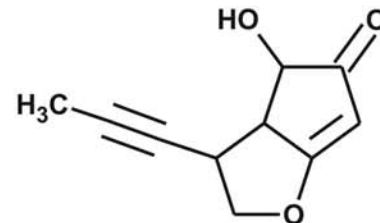
typical van der Waals volume of $C_{10}H_{10}O_3$: $160 \text{ \AA}^3 \xrightarrow{V^{-1}} 0.63 * 10^{28} \text{ m}^{-3}$



153.4 \AA^3



164.3 \AA^3

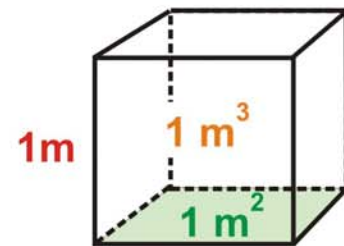


$C_{10}H_{10}O_3$

160.3 \AA^3

“diversity index” of $C_{10}H_{10}O_3$ molecules in one cubic meter

$$1.1133 * 10^7 * 0.63 * 10^{28} \text{ m}^{-3} = 7.01 * 10^{34}$$



introduction

$$h\nu(^{133}\text{Cs}) = 9.192.631.770 \text{ Hz}$$

high precision frequency measurements are
manna from heaven for **molecular-level
resolution structural analysis**

Nobel price 2005 (MPI Munich)

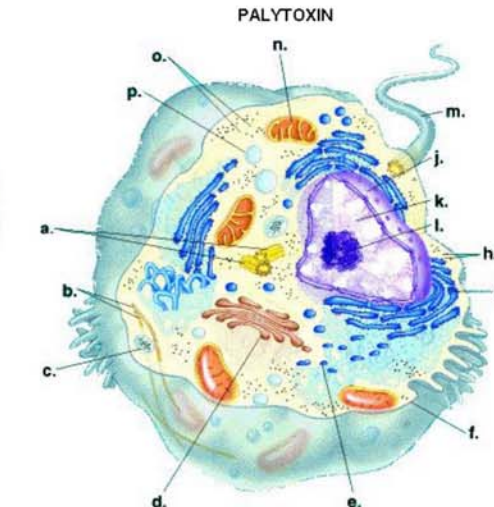
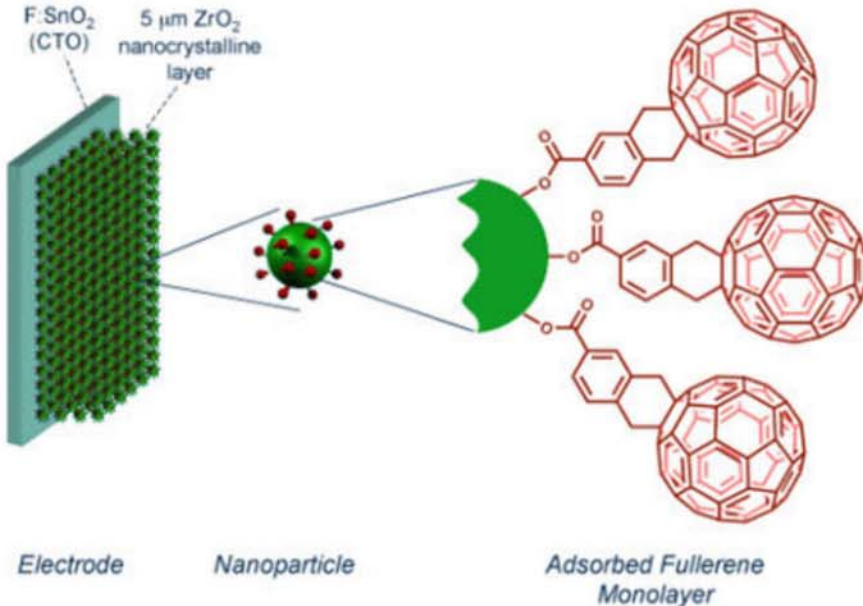
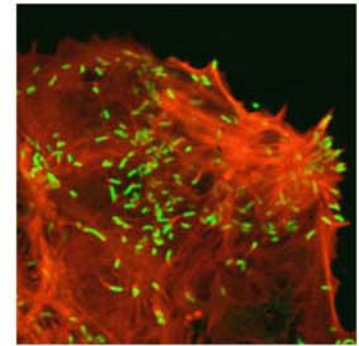
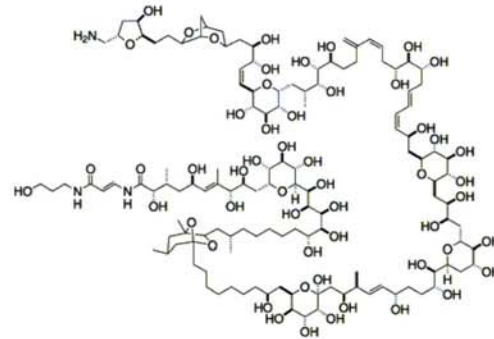
today:



*molecular-level perception of natural
ecosystems requires educated handling
of huge data sets in excess of 10^9 “pixels”*

complex systems

any "**non-repetitive**" *non-protein*
natural or synthetic material



aspects of molecular complexity

atomic signature
molecular signature

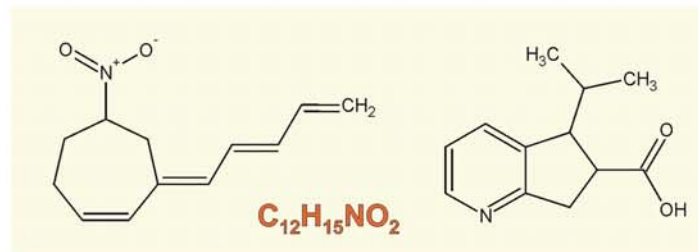
compositional



molecular formula

FTICR mass spectrometry

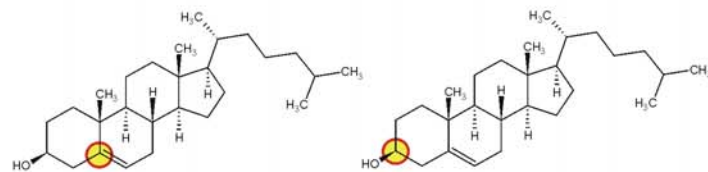
isomeric structures



atomic connectivities and spatial orientation

NMR spectroscopy

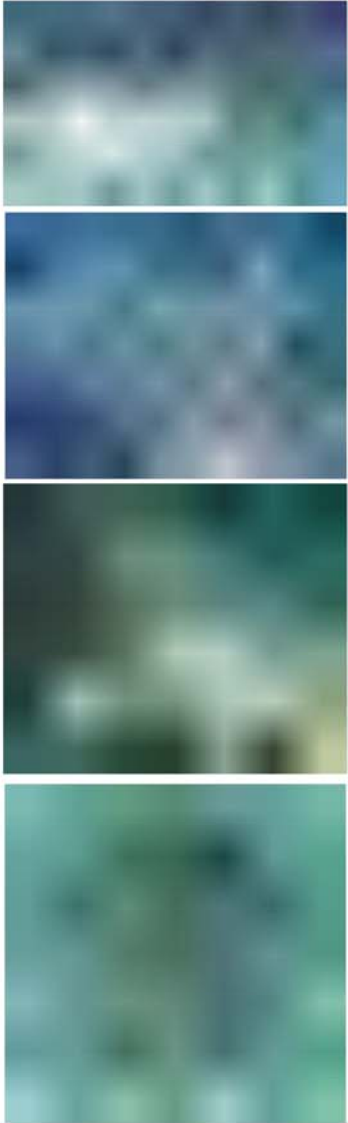
isotopomers



positions of (stable) isotopes within molecules

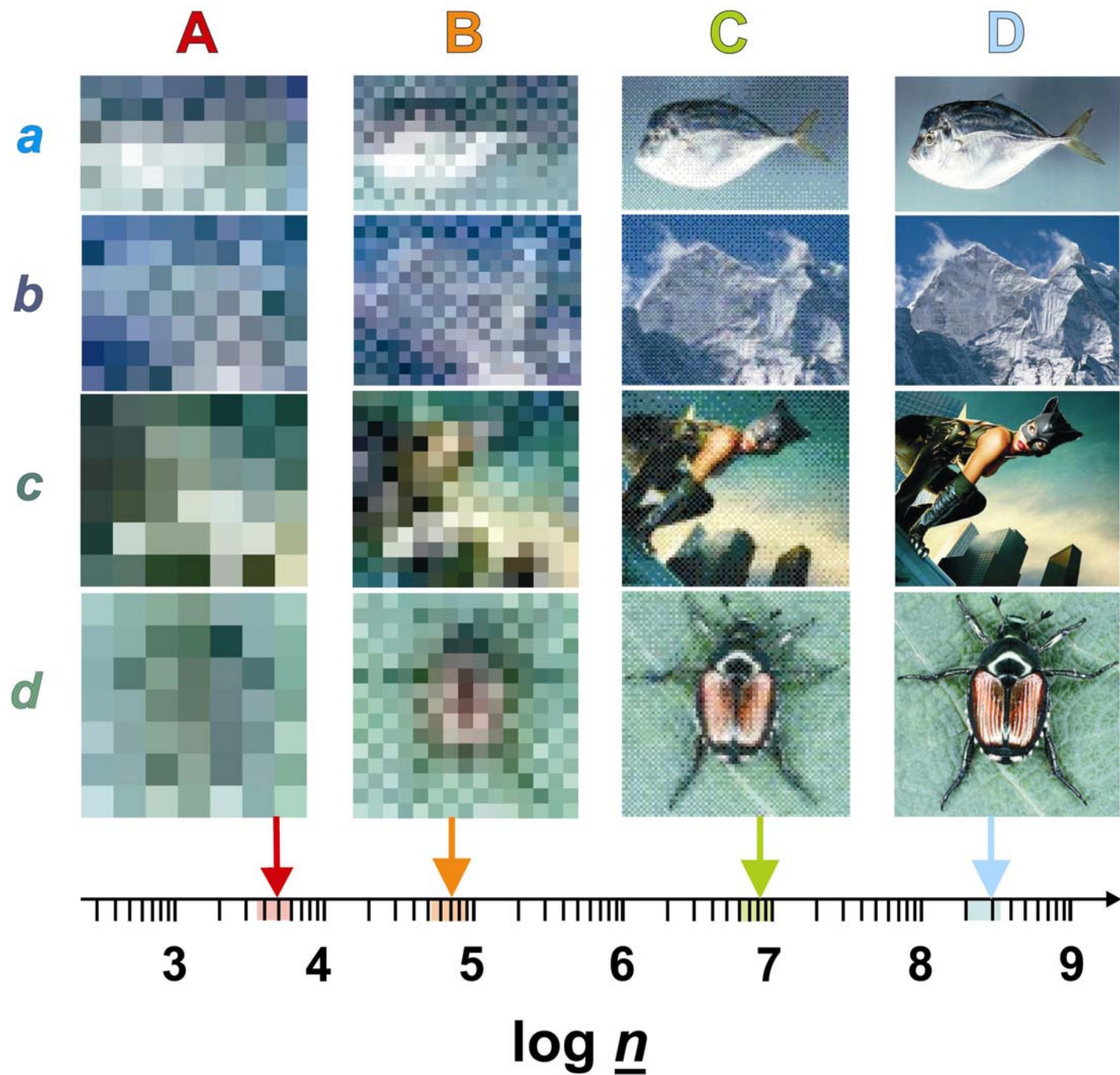
NMR spectroscopy

intrinsic averaging in low resolution analytical characterization of NOM

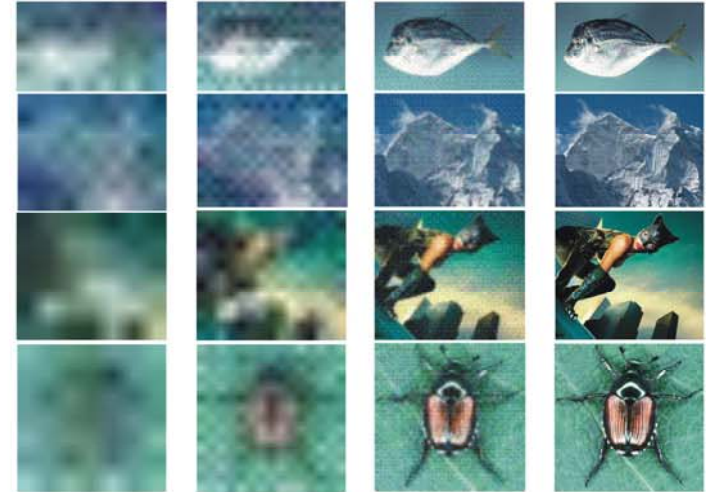
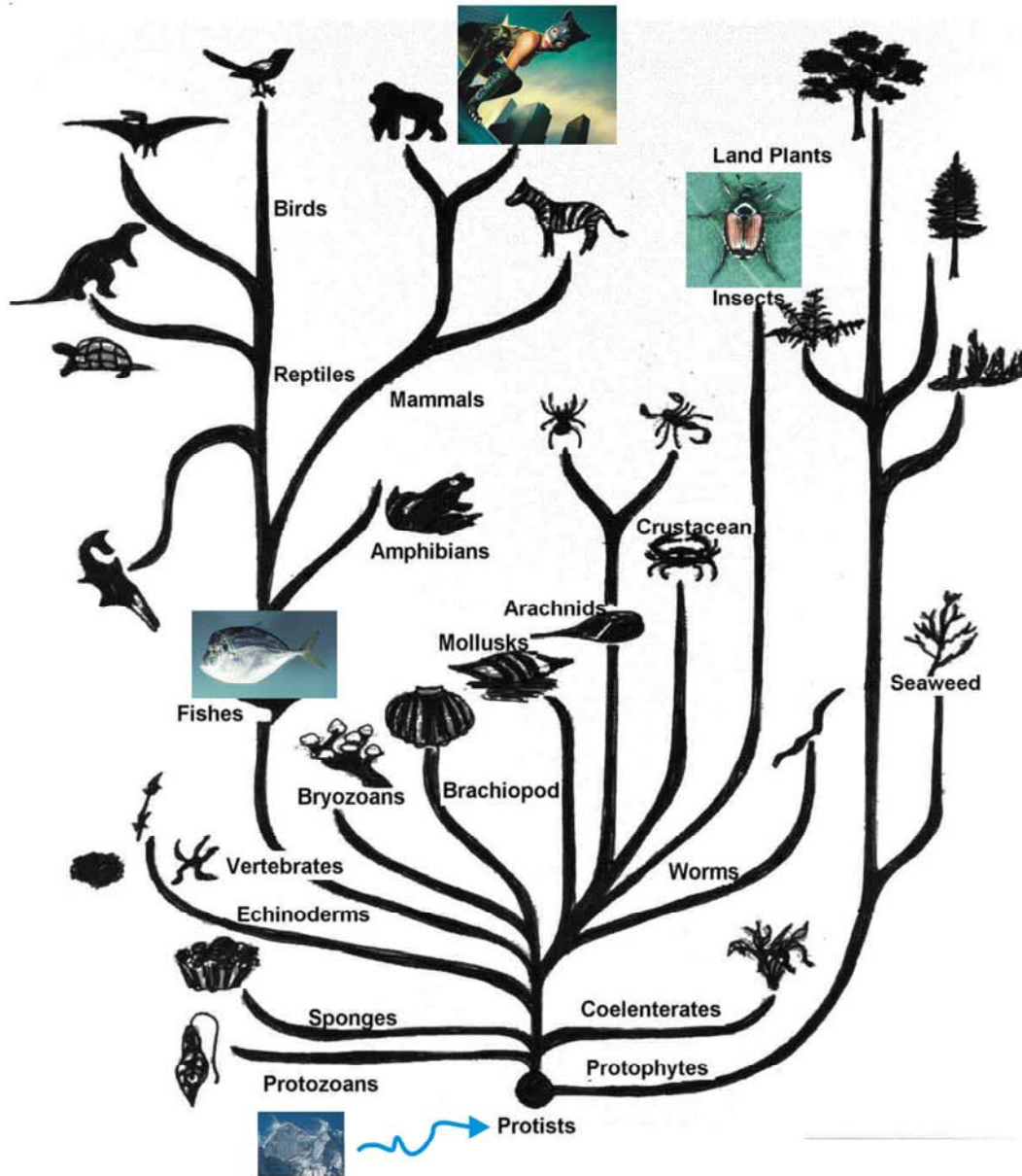


UV/VIS titration
elemental analysis

why do all NOM
appear to be
so similar ?



sufficient resolution allows meaningful analysis of processes



introduction

$$h\nu(^{133}\text{Cs}) = 9.192.631.770 \text{ Hz}$$

high precision frequency measurements are
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Nobel price 2005 (MPI Munich)

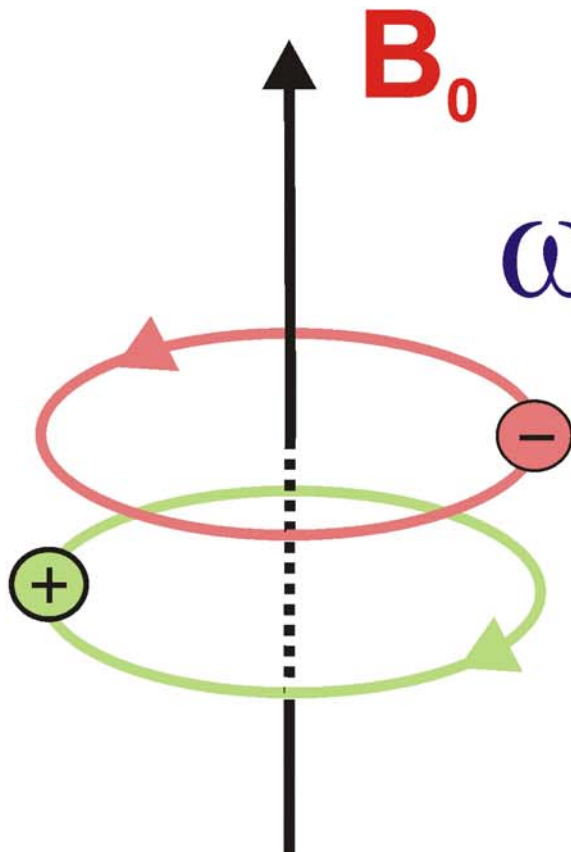
today:



*molecular-level perception of natural
ecosystems requires educated handling
of huge data sets in excess of 10^9 “pixels”*

FTICR mass spectrometry / molecular process

orbital frequency ν_c directly relates to m/z

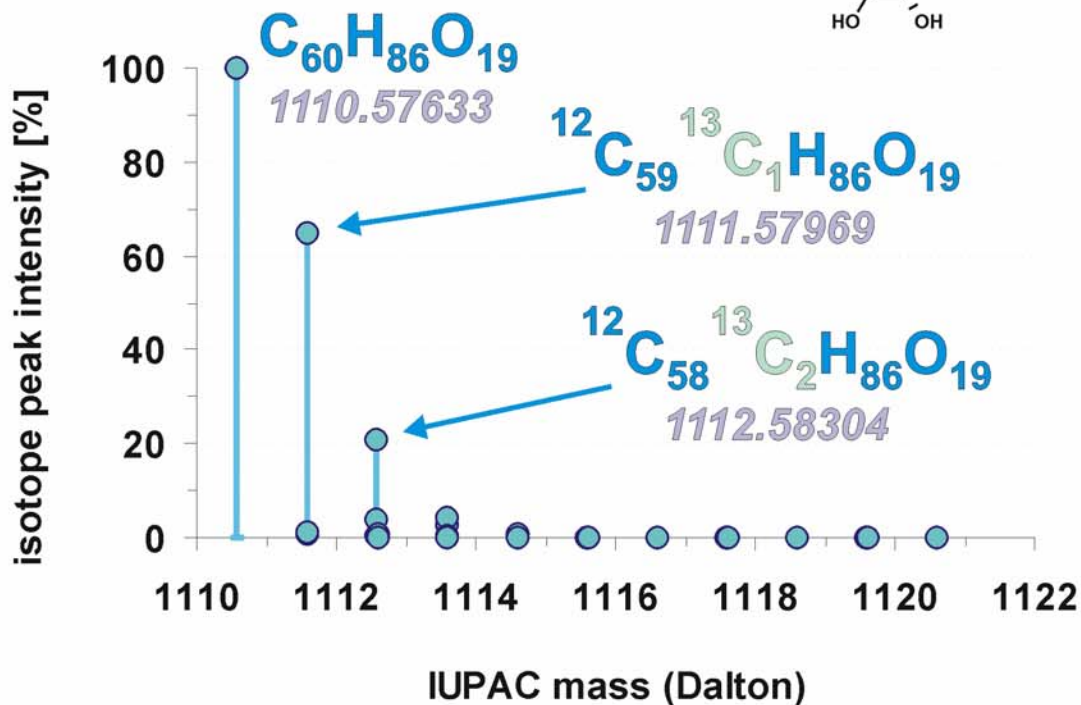
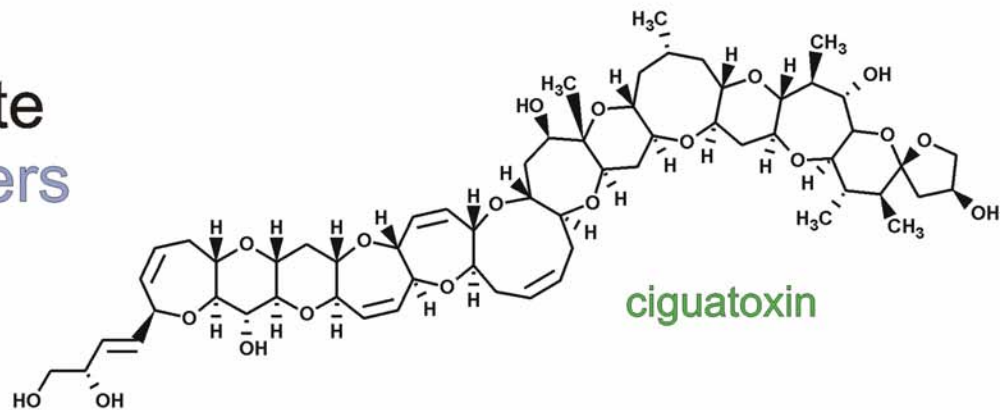


$$\omega_c = z \cdot B_0 / m$$

$$\omega_c = \frac{\nu_c}{2\pi} = \frac{1.535611 B_0 \cdot z}{m}$$

FTICR mass spectrometry / molecular process

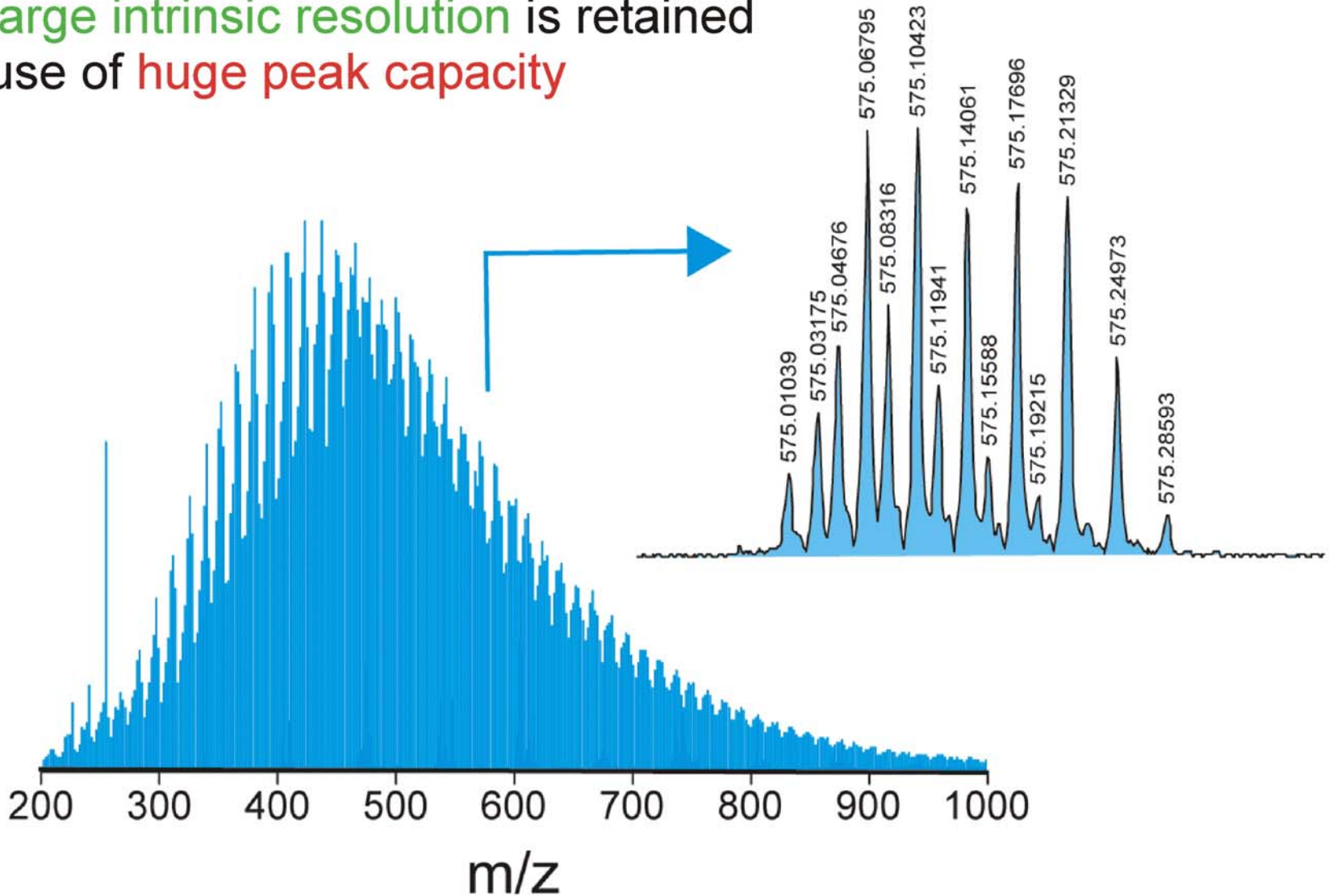
FTICR mass spectra indicate
molecular ions of isotopomers



IUPAC mass of ciguatoxin
 $C_{60}H_{86}O_{19} = 1111.313$ Da
(reflecting average isotopic composition)

FTICR mass spectrometry / complex system

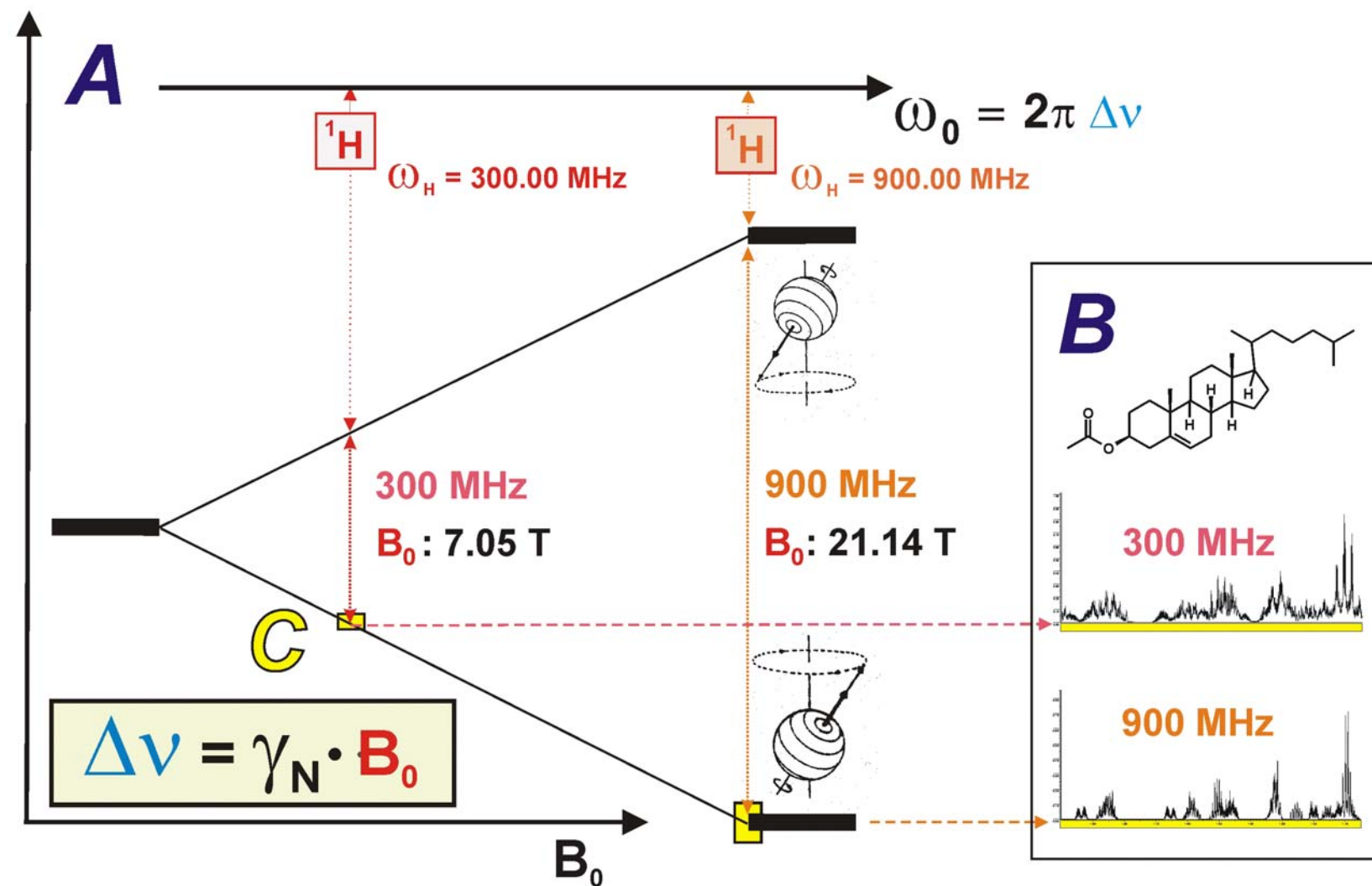
very **large intrinsic resolution** is retained
because of **huge peak capacity**



NMR spectroscopy / atomic process

transitions among individual atomic energy levels

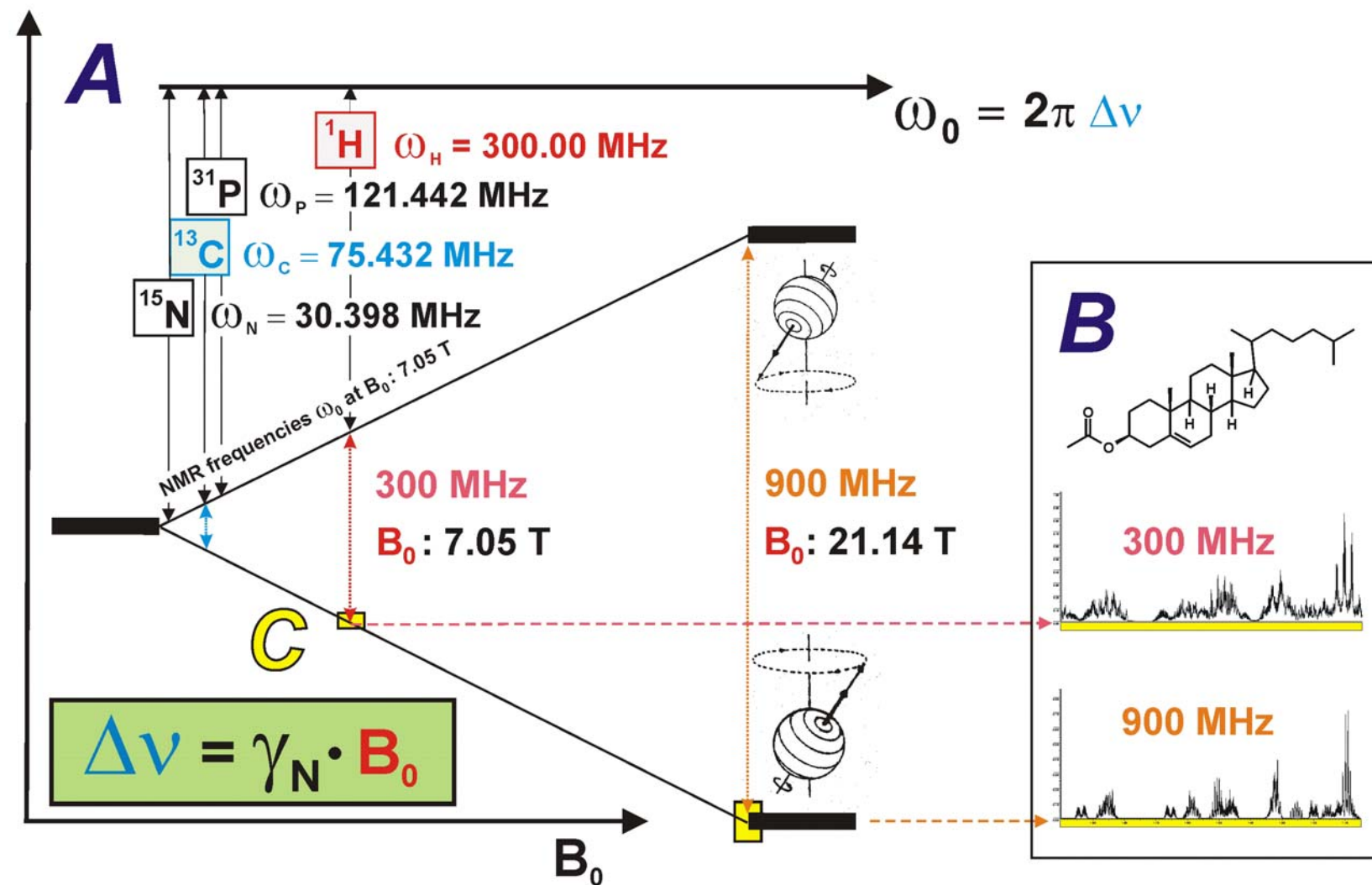
$$\Delta E = h \Delta \nu$$



NMR spectroscopy / atomic process

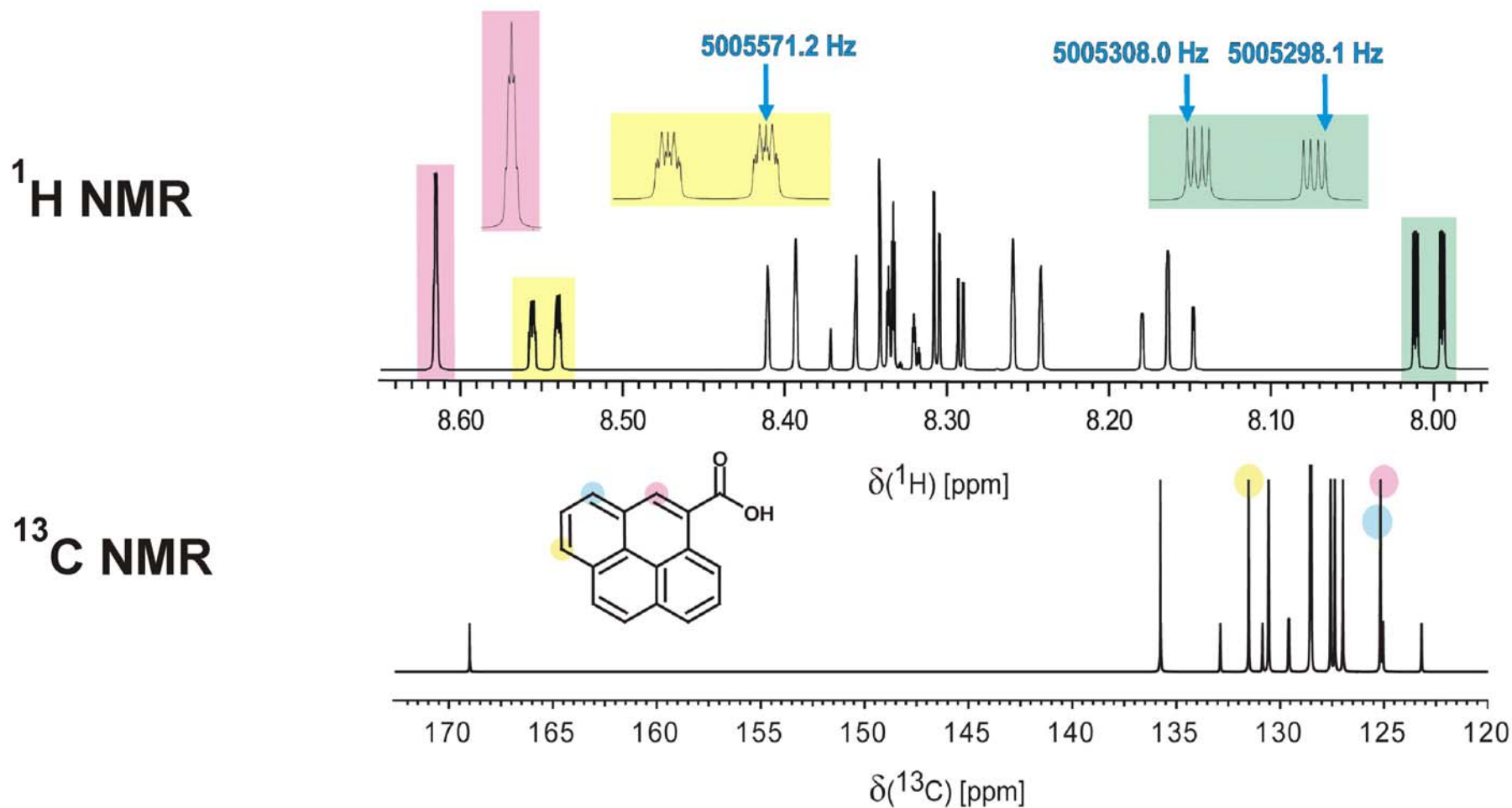
transitions among individual atomic energy levels

$$\Delta E = h \Delta \nu$$



NMR spectroscopy / molecular process

atomic signatures in molecules allow
unambiguous assembly of (isomeric) structures

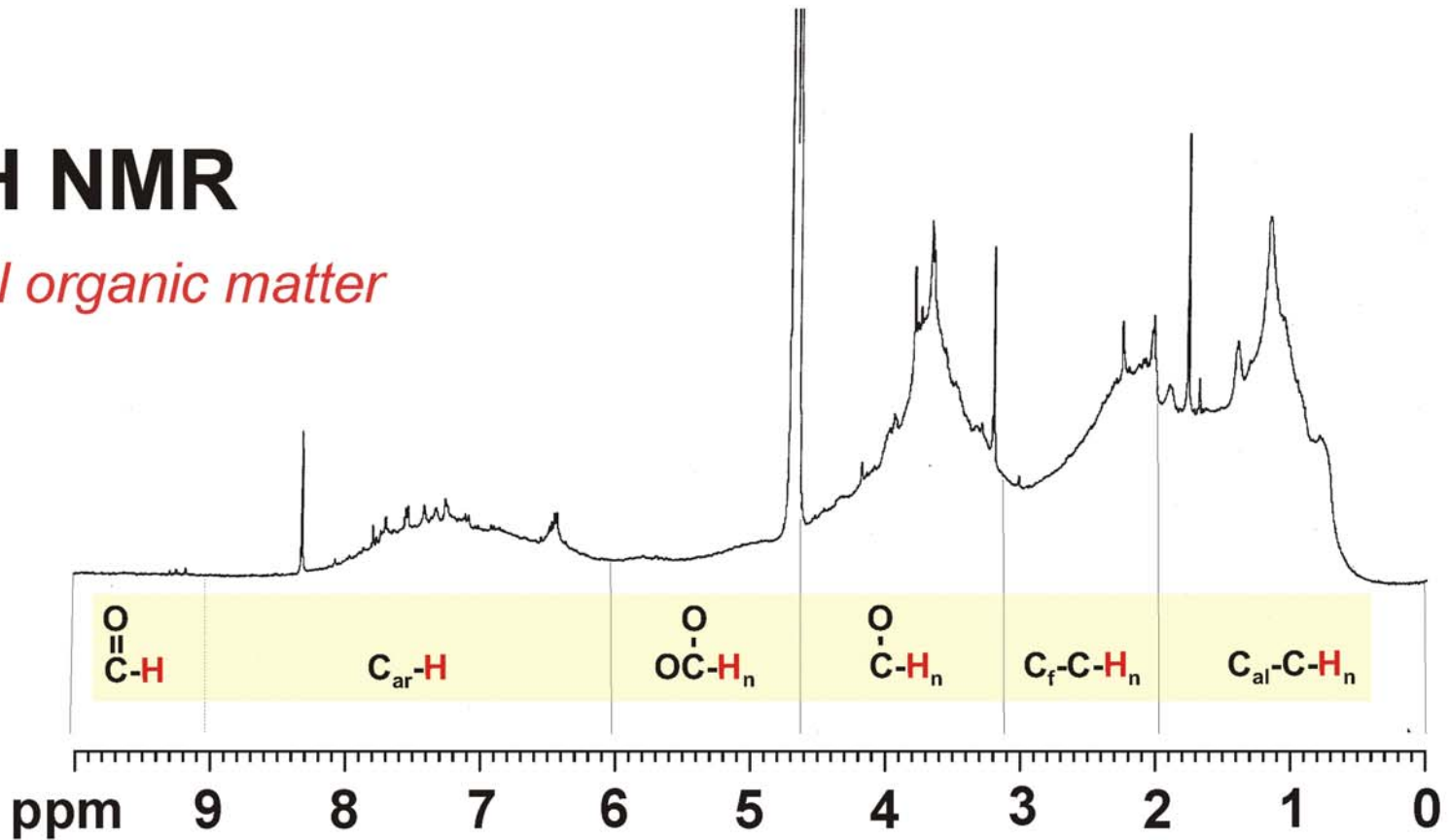


NMR spectroscopy / complex systems

NMR section integrals indicate fundamental substructure regimes

massive overlap interferes with resolution

^1H NMR
natural organic matter



complex systems

high-precision frequency measurements are at the core of the *two most insightful techniques*, which provide **molecular resolution** information about organic molecules

mass spectrometry

data reduced



NMR spectroscopy

information-rich



$$e^{i\pi} = -1$$

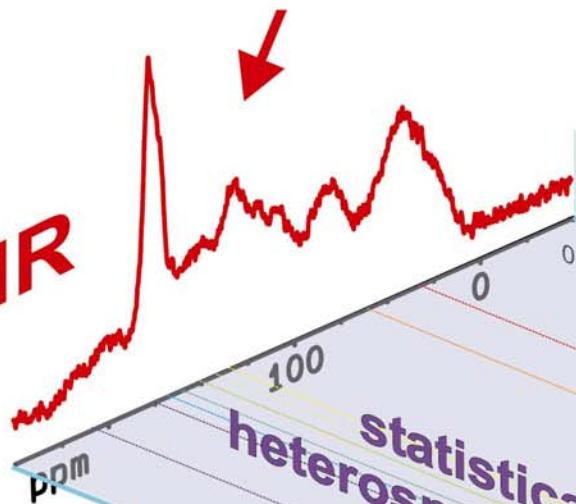
the total space of
molecular structures

10^{60-200}

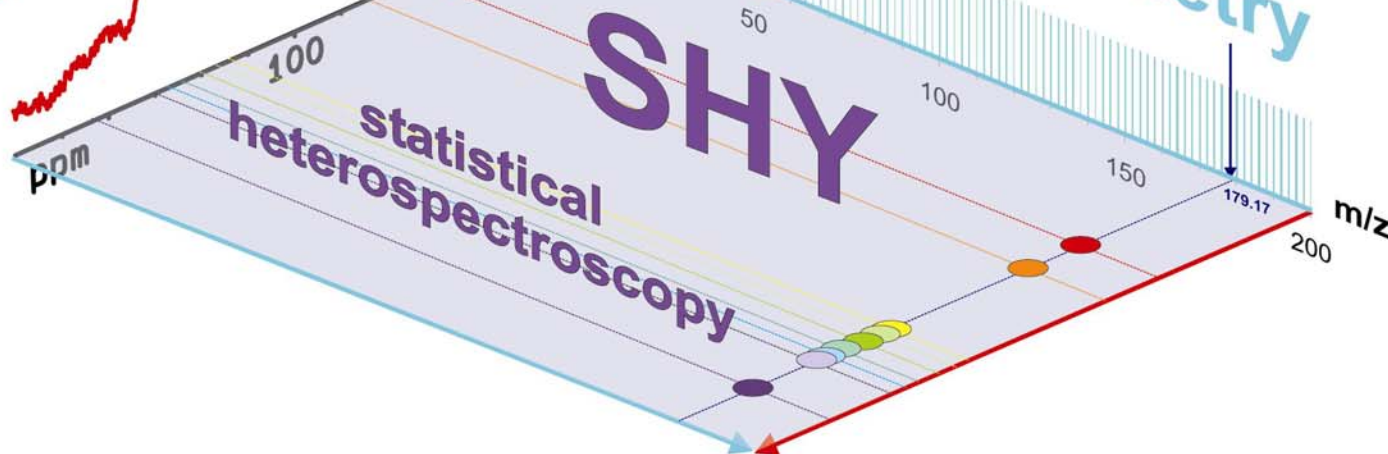
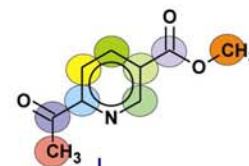
*isotope-specific
projection of
molecular environments*

*isomer-filtered projection results in
the compositional space*

NMR

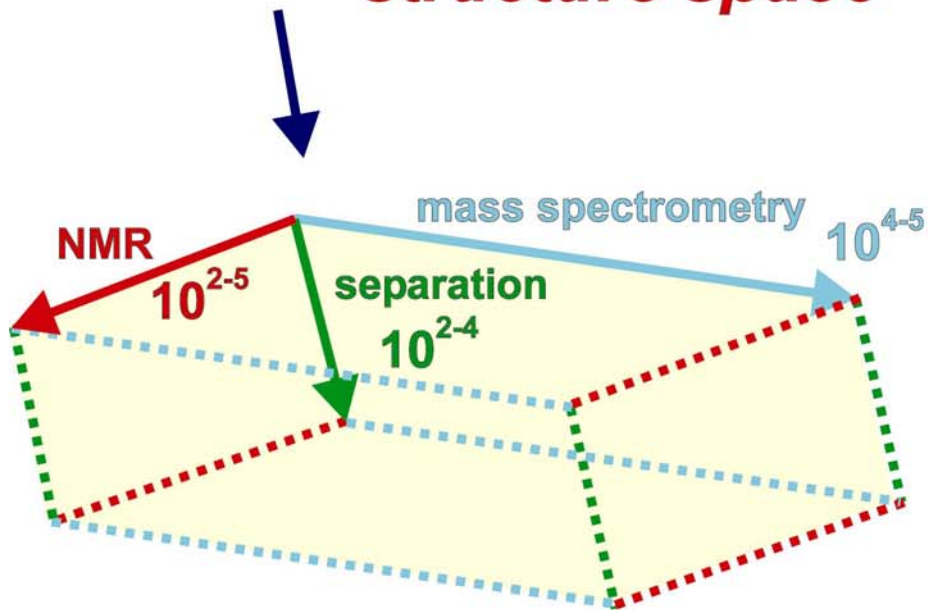


mass spectrometry



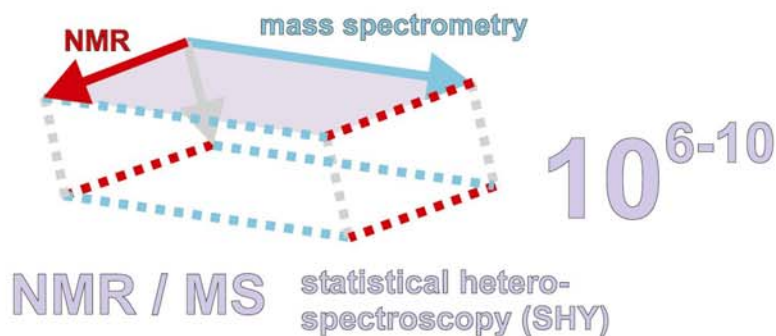
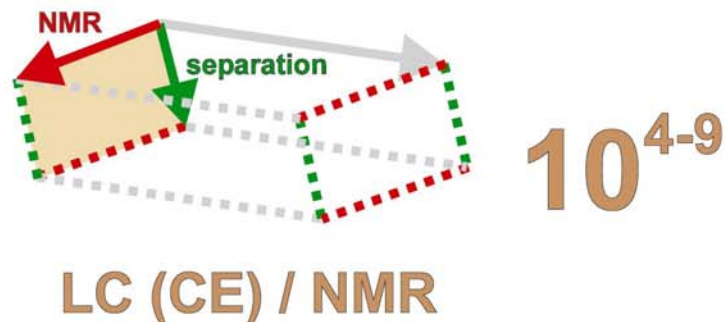
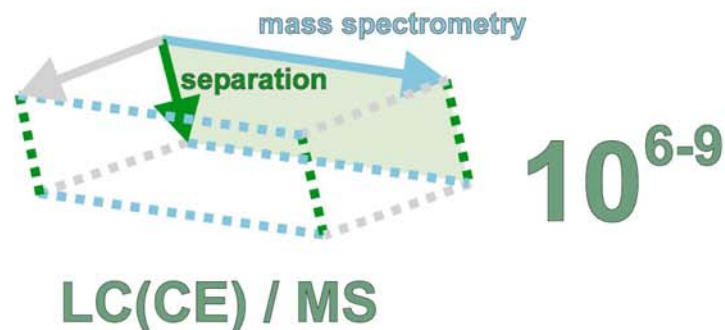
10^{60-200}

structure space

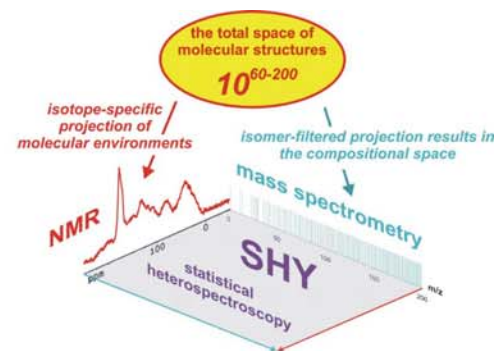
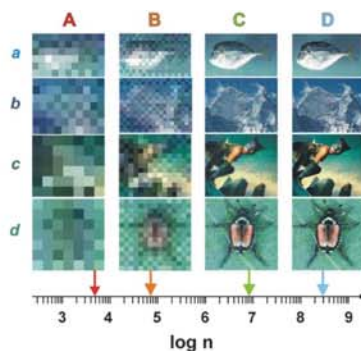


significant (analytical)

bucket space: 10^{8-14}



...what is the utility of these novel tools....?



High-resolution organic structural spectroscopy already has advanced crucial **paradigm shifts** in structural characterization of **natural organic matter (NOM)**

Hertkorn et al., Anal. Bioanal. Chem. 2007, 389, 1311-1327

- discovery of **carboxyl-rich alicyclic molecules (CRAM)** as **major constituents of NOM**

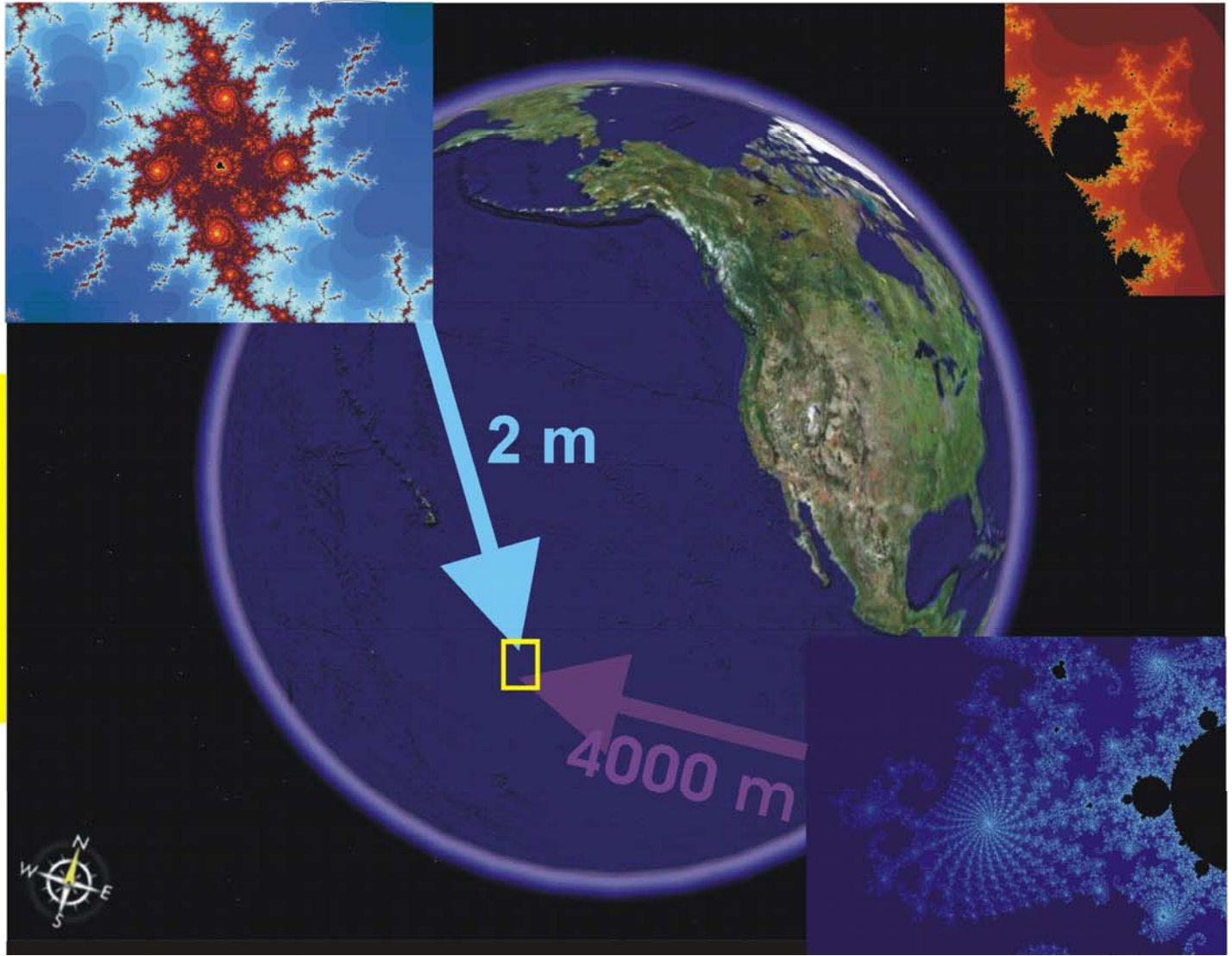
Hertkorn et al., GCA 2006, 71, 2995-3010

Lam et al., Environ. Sci. Technol. 2007, 54, 8240-8247

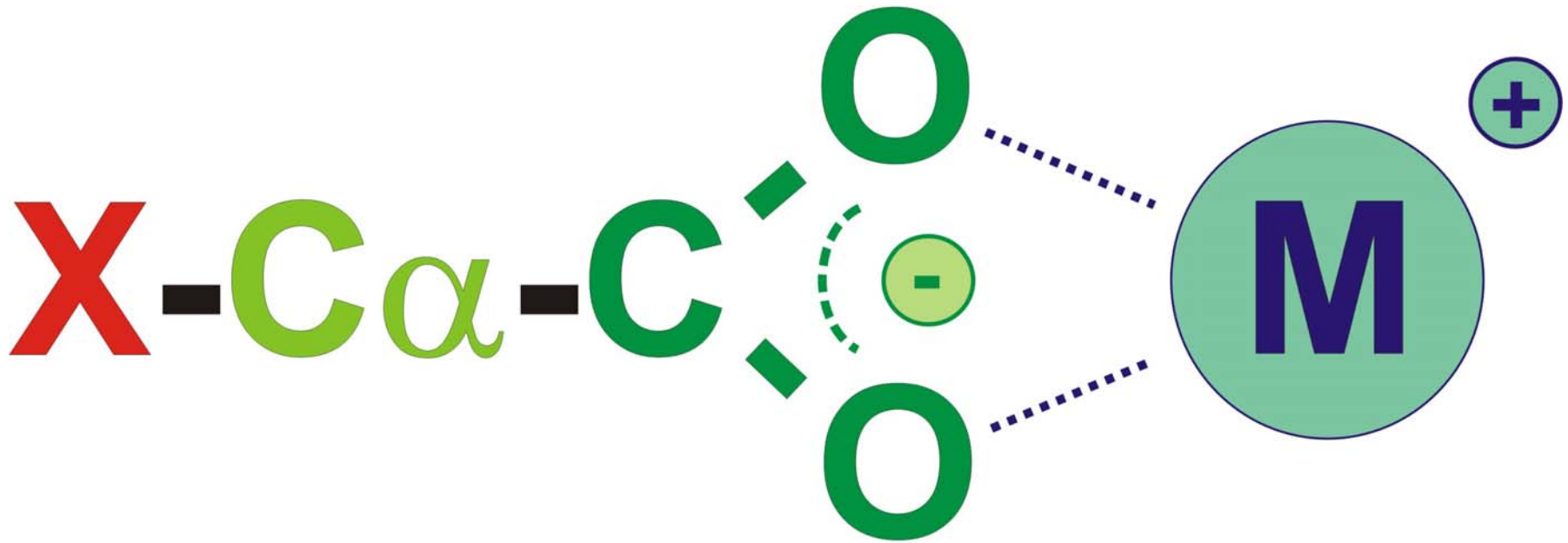
- large-scale **molecular turnover** of NOM on **short time scales**

Einsiedl et al., GCA, 2007, 71, 5474-5482

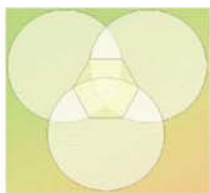
CRAM (carboxyl-rich alicyclic molecules) have been first identified in the surface and deep Pacific ocean; there is good reason to postulate **CRAM** as a major constituent of any **NOM**



aliphatic polycarboxylic acid

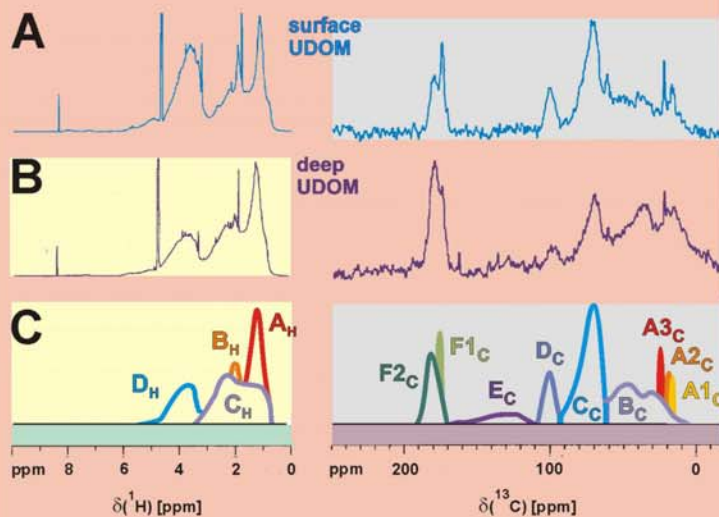


carboxyl-rich **alicyclic** molecules CRAM

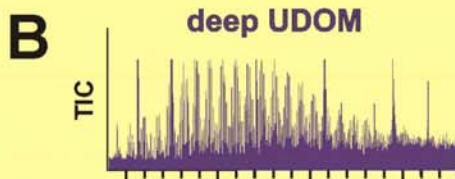
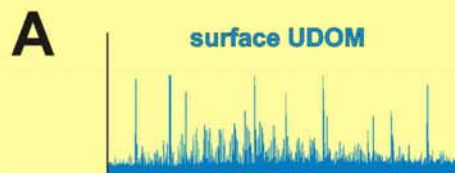


complementary techniques have led to the identification of **CRAM** (carboxylic-rich alicyclic molecules) in **NOM**

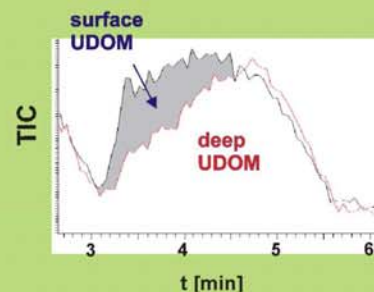
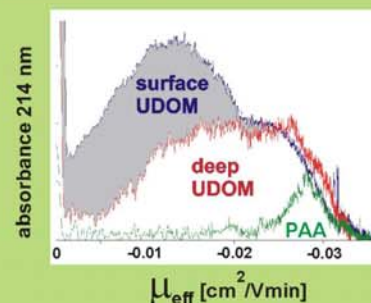
NMR quantification



ACS_@de_compl_method.cdr

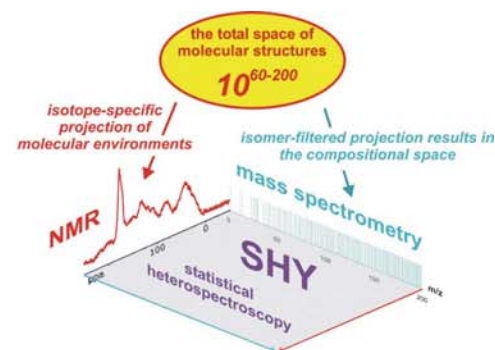
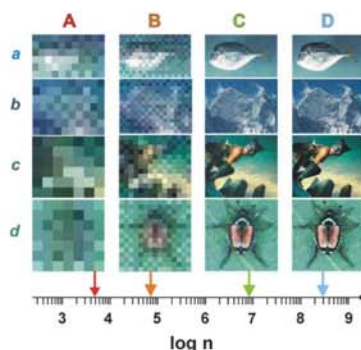


mass spectrometry resolution



capillary electrophoresis validation

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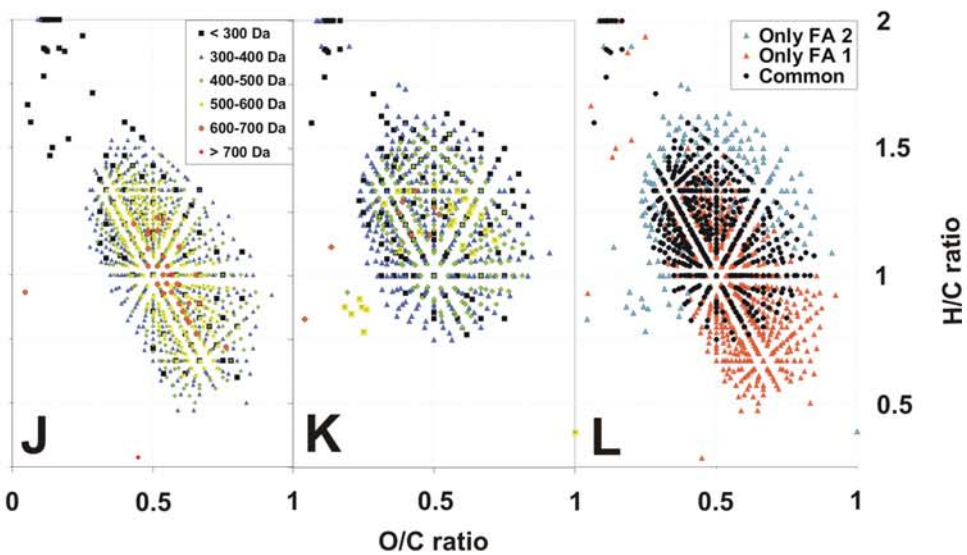
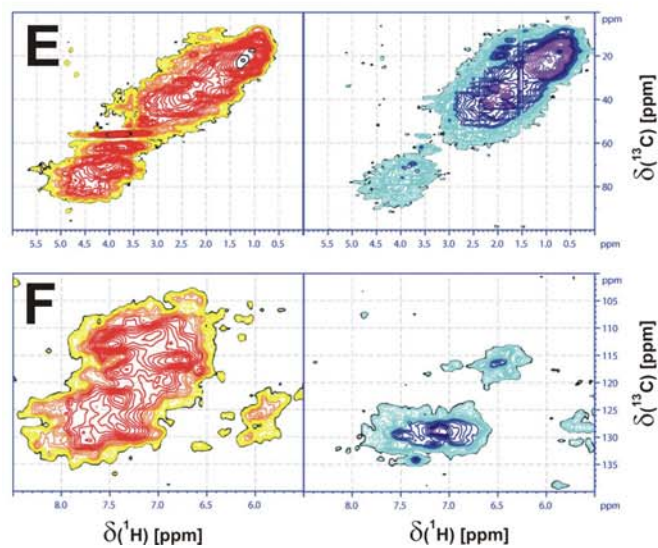
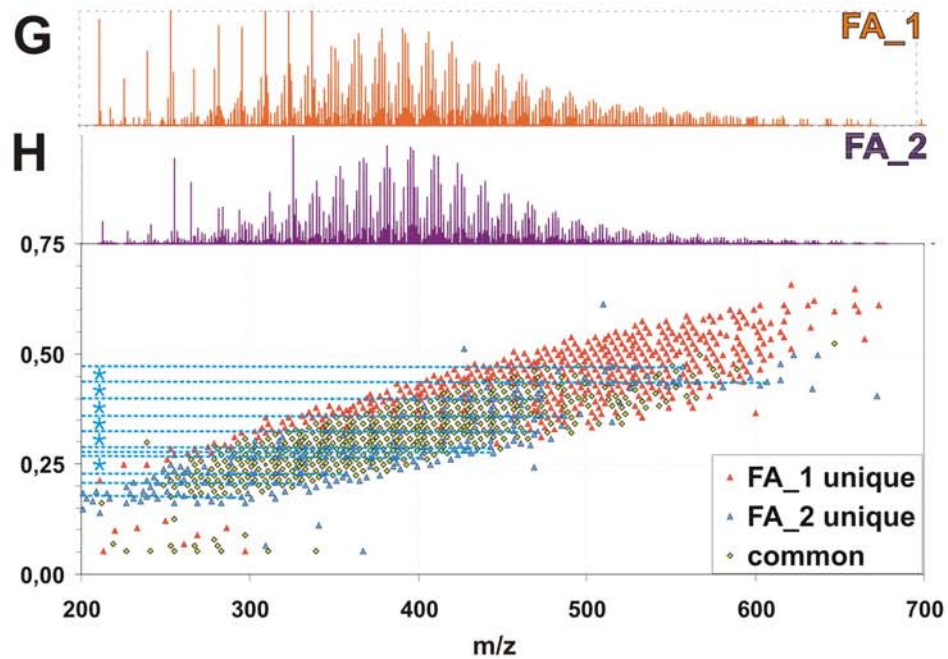
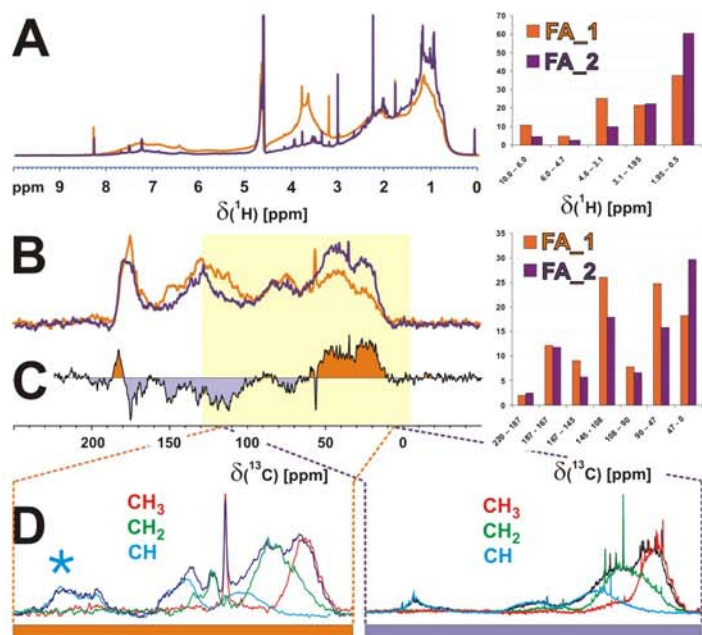
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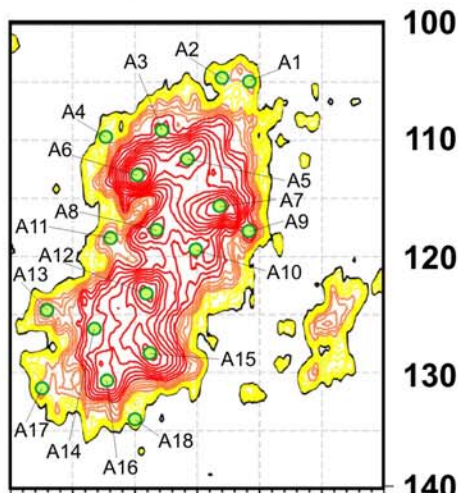
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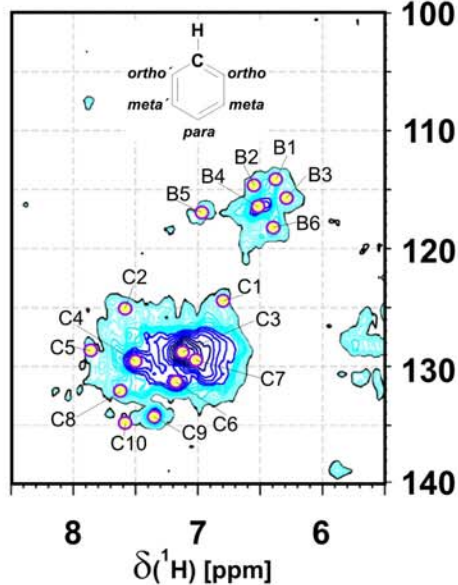
evolution of groundwater during 60 years

SPARIA analysis (**S**ubstitution **P**atterns in **A**romatic **R**ings by **I**crement **A**nalysis)

FA1
0 y



FA2
60y



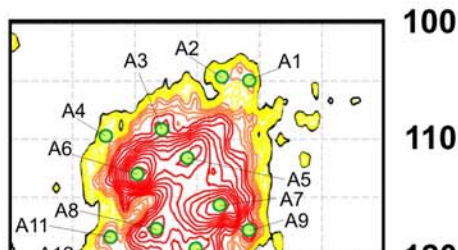
FA1									
peak	$\delta(^1\text{H})$ [ppm]	$\delta(^{13}\text{C})$ [ppm]	ortho	meta	para	meta'	ortho'	percent	ne
A1	6.57	104.6						27	67
A2	6.79	104.4						45	20
A3	7.28	108.7						37	45
A4	7.73	109.5						50	2
A5	7.08	111.4						32	38
A6	7.49	112.6						41	59
A7	6.82	115.5						19	52
A8	7.33	117.4						17	59
A9	6.57	117.6						33	36
A10	7.13	120.2						31	78
A11	7.39	118.2						17	78
A12	7.42	122.8						14	114
A13	8.21	124.2						25	75
A14	7.83	126.0						14	122
A15	7.38	128.0						28	43
A16	7.74	130.3						33	812
A17	8.25	131.1						41	87
A18	7.50	133.6						52	48
FA2									
peak	$\delta(^1\text{H})$ [ppm]	$\delta(^{13}\text{C})$ [ppm]	ortho	meta	para	meta'	ortho'	percent	ne
B1	6.39	113.9						36	33
B2	6.53	114.6						36	29
B3	6.29	115.3						100	10
B4	6.52	116.0						23	26
B5	6.97	116.6						24	104
B6	6.39	117.9						36	14
C1	6.79	124.1						33	30
C2	7.58	124.8						21	63
C3	7.12	128.3						19	36
C4	7.51	129.1						29	59
C5	7.86	128.3						28	97
C6	7.17	131.0						28	42
C7	7.05	128.9						43	30
C8	7.62	131.6						33	93
C9	7.34	133.8						36	56
C10	7.59	134.4						87	15

	aromatic substitution, composed of hydrogen
	aromatic substitution, composed of neutral carbon substituents
	aromatic substitution, composed of electron-withdrawing carbonyl derivative substituents
	aromatic substitution, composed of electron-donating oxygen substituents

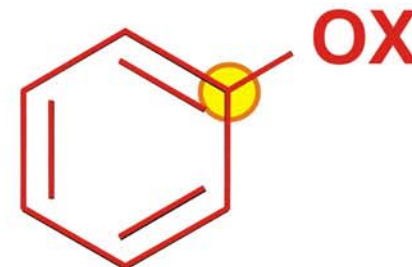
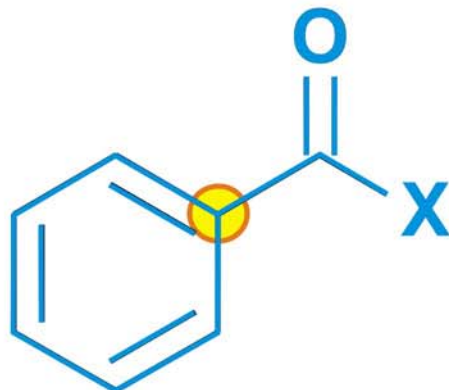
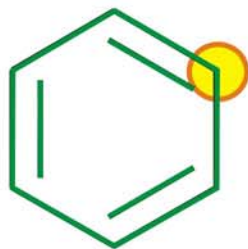
evolution of groundwater during 60 years

SPARIA analysis (Substitution Patterns in Aromatic Rings by Icrement Analysis)

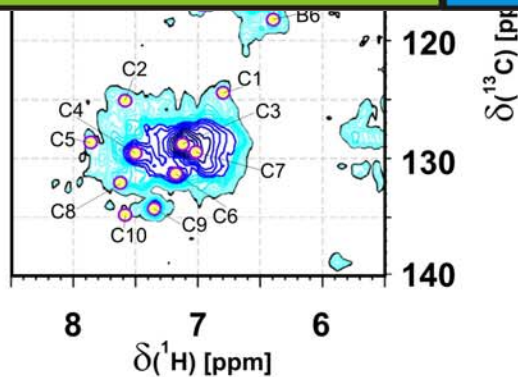
FA1



FA1									
peak	$\delta(^1\text{H})$ [ppm]	$\delta(^{13}\text{C})$ [ppm]	ortho	meta	para	meta'	ortho'	percent	ne
A1	6.57	104.6						27	67
A2	6.79	104.4						45	20
A3	7.28	108.7						37	45
A4	7.73	109.5						50	2
A5	7.08	111.4						32	38
A6	7.49	112.6						41	59



FA2
60y



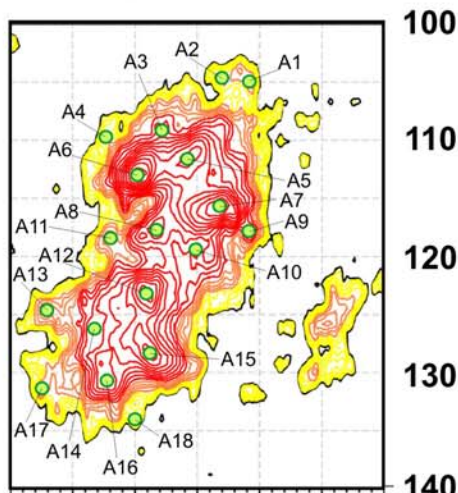
B6	6.39	117.9						36	14
C1	6.79	124.1						33	30
C2	7.58	124.8						21	63
C3	7.12	128.3						19	36
C4	7.51	129.1						29	59
C5	7.86	128.3						28	97
C6	7.17	131.0						28	42
C7	7.05	128.9						43	30
C8	7.62	131.6						33	93
C9	7.34	133.8						36	56
C10	7.59	134.4						87	15

	aromatic substitution, composed of hydrogen
	aromatic substitution, composed of neutral carbon substituents
	aromatic substitution, composed of electron-withdrawing carbonyl derivative substituents
	aromatic substitution, composed of electron-donating oxygen substituents

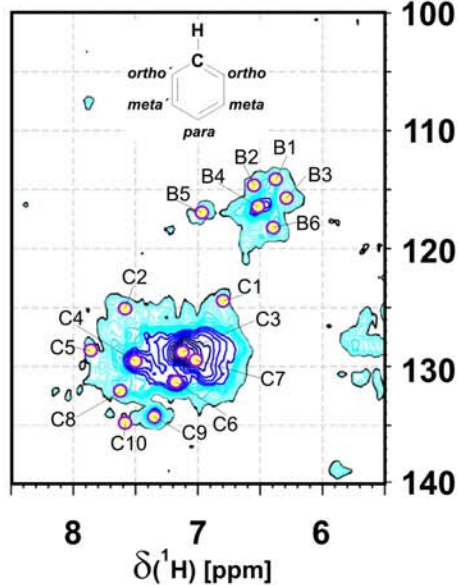
evolution of groundwater during 60 years

SPARIA analysis (**S**ubstitution **P**atterns in **A**romatic **R**ings by **I**crement **A**nalysis)

FA1
0 y



FA2
60y



FA1									
peak	$\delta(^1\text{H})$ [ppm]	$\delta(^{13}\text{C})$ [ppm]	ortho	meta	para	meta'	ortho'	percent	ne
A1	6.57	104.6						27	67
A2	6.79	104.4						45	20
A3	7.28	108.7						37	45
A4	7.73	109.5						50	2
A5	7.08	111.4						32	38
A6	7.49	112.6						41	59
A7	6.82	115.5						19	52
A8	7.33	117.4						17	59
A9	6.57	117.6						33	36
A10	7.13	120.2						31	78
A11	7.39	118.2						17	78
A12	7.42	122.8						14	114
A13	8.21	124.2						25	75
A14	7.83	126.0						14	122
A15	7.38	128.0						28	43
A16	7.74	130.3						33	812
A17	8.25	131.1						41	87
A18	7.50	133.6						52	48
FA2									
peak	$\delta(^1\text{H})$ [ppm]	$\delta(^{13}\text{C})$ [ppm]	ortho	meta	para	meta'	ortho'	percent	ne
B1	6.39	113.9						36	33
B2	6.53	114.6						36	29
B3	6.29	115.3						100	10
B4	6.52	116.0						23	26
B5	6.97	116.6						24	104
B6	6.39	117.9						36	14
C1	6.79	124.1						33	30
C2	7.58	124.8						21	63
C3	7.12	128.3						19	36
C4	7.51	129.1						29	59
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	aromatic substitution, composed of hydrogen
	aromatic substitution, composed of neutral carbon substituents
	aromatic substitution, composed of electron-withdrawing carbonyl derivative substituents
	aromatic substitution, composed of electron-donating oxygen substituents

conclusion

**natural organic matter (NOM) is
NOT refractory
on the molecular level**

the current **perception** of **NOM** appearing refractory is mainly an inevitable consequence of **intrinsic averaging** produced by **low-resolution analytical methods**

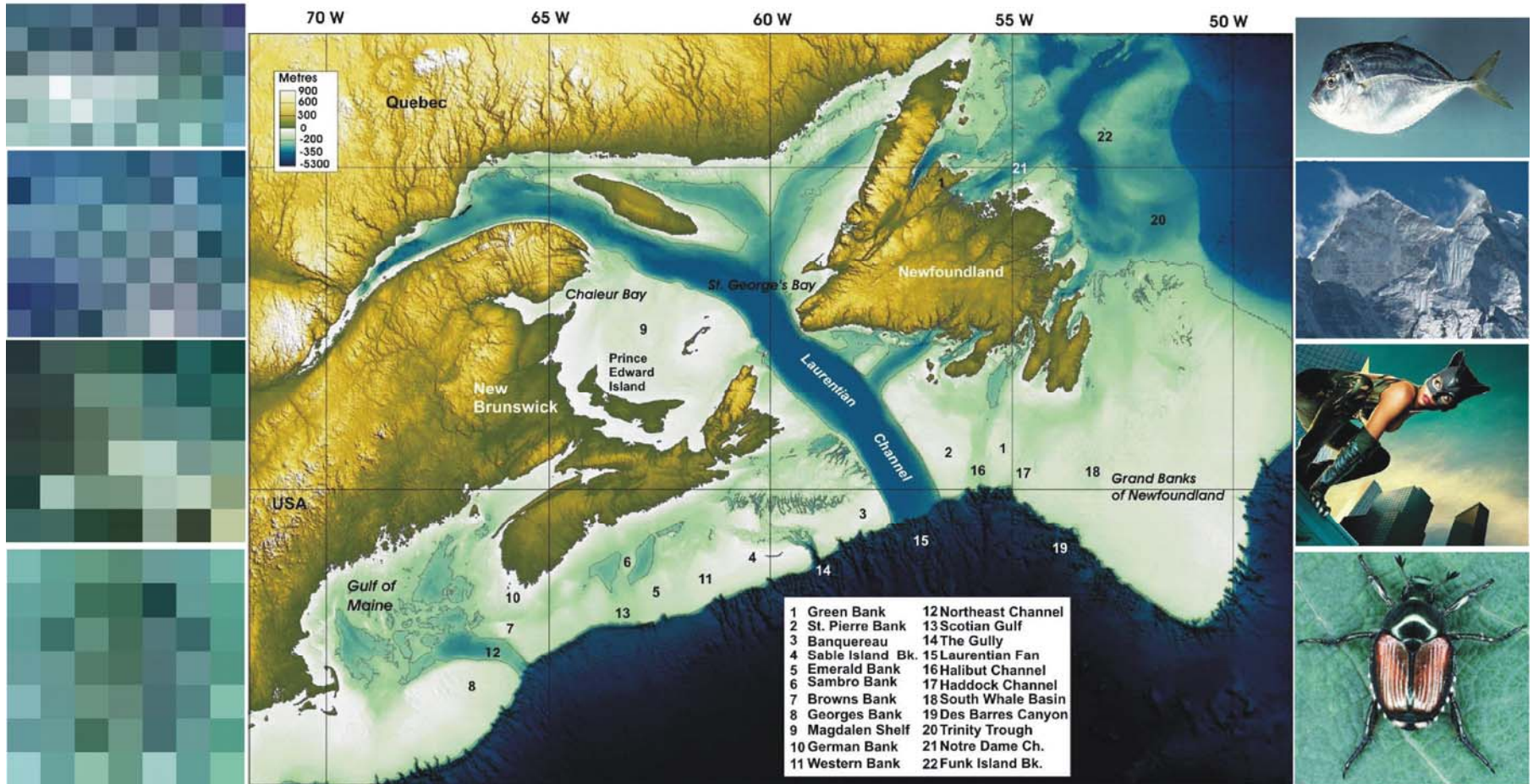
.....*and most often likely false*

ten-year-vision:

authentic molecular representation of complex natural systems

2005

2015

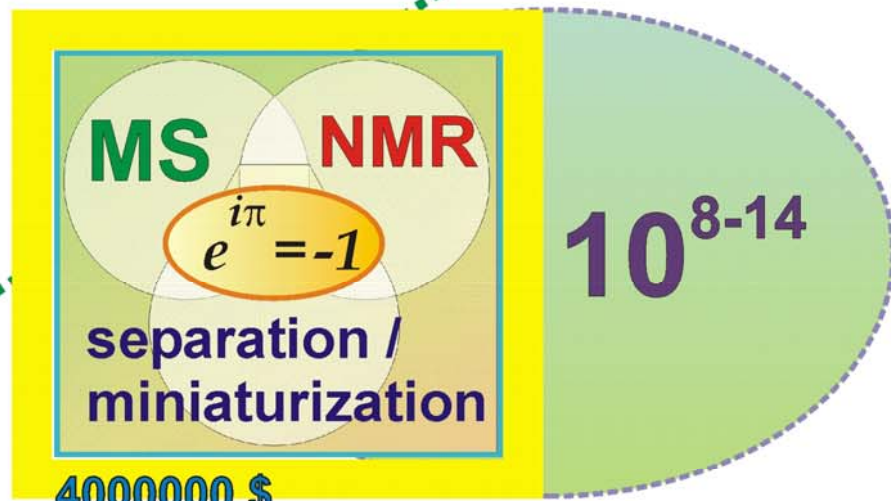


peeking beyond the compositional space



chemistry

10^{60-200}

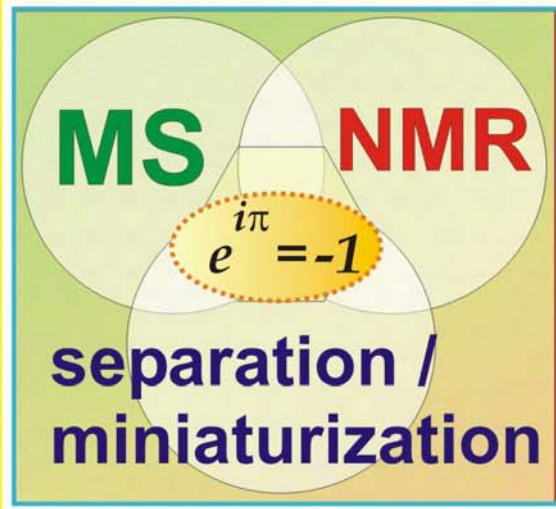


10^{4-5}

FTICR mass spectrometry

integrated systems analysis
complementary
*.omics approach

molecular-level structural analysis of non-repetitive complex unknowns



40000000 \$



complementary techniques, aspects, **brains**

NMR

quantification

mass spectrometry

resolution

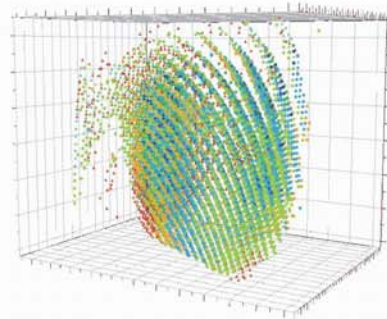
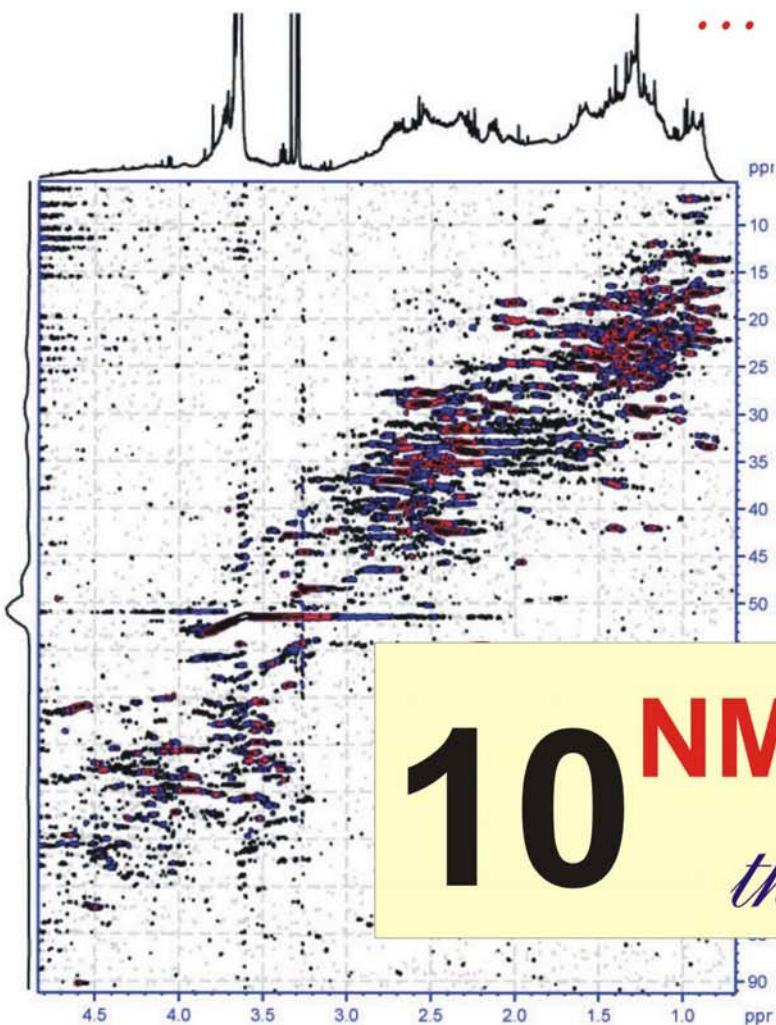
high-performance separation

validation

integrated mathematical systems analytics

conclusion

... I want to stay in NMR



10 NMR+FTMS+sep+ $e^{i\pi} = -1$
the unbeatable combination

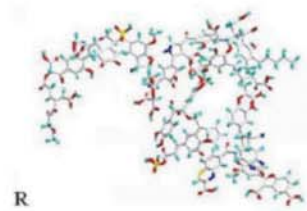
acknowledgements

S. Schulte-Hostede

A. Kettrup

for generous and wonderful freedom

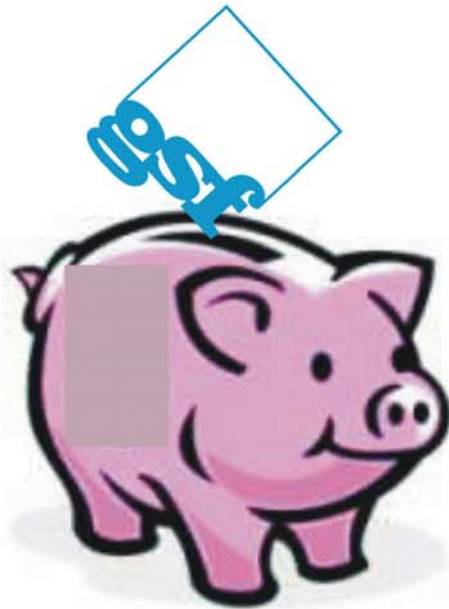
our worldwide partners



support: Alexander von Humboldt-Stiftung,

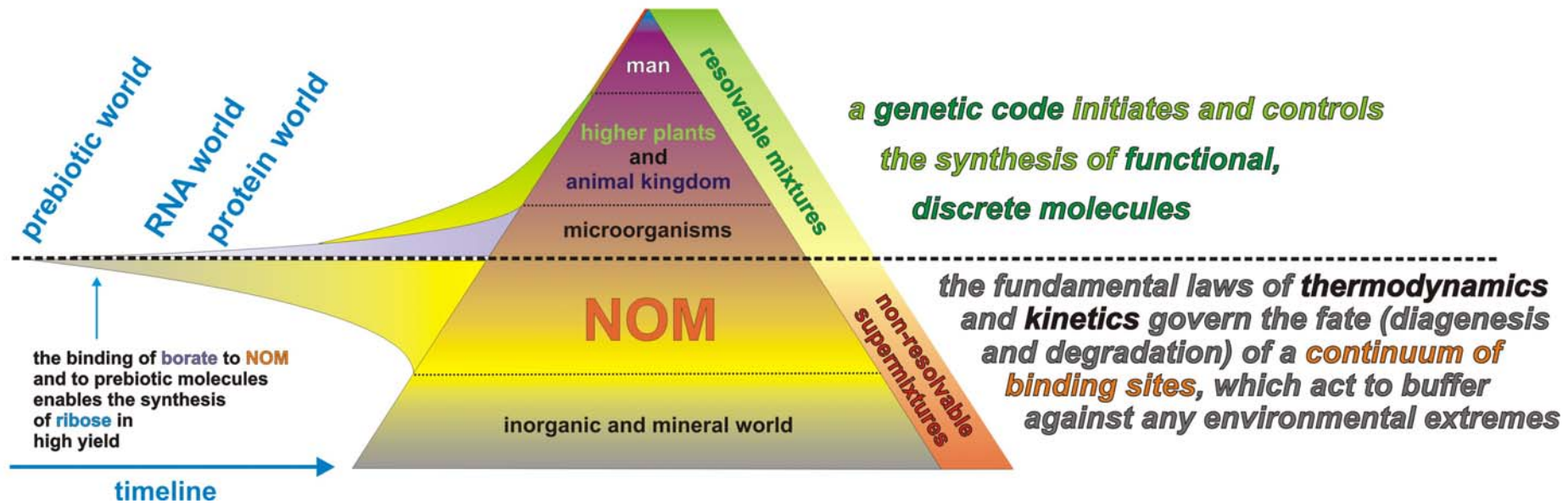
DFG, EU, DAAD

NSF, GIF, GSF

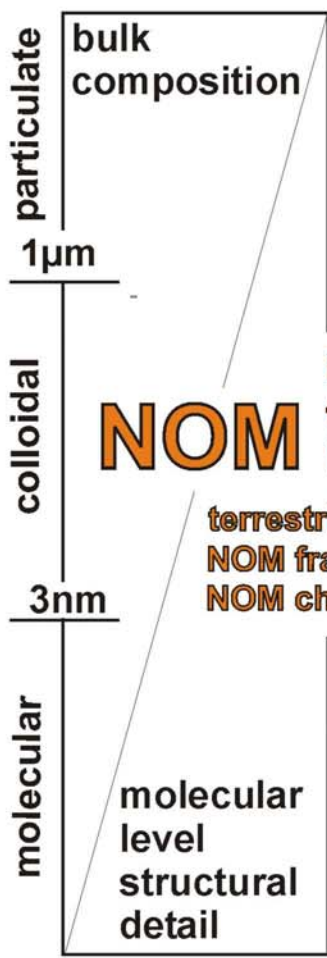
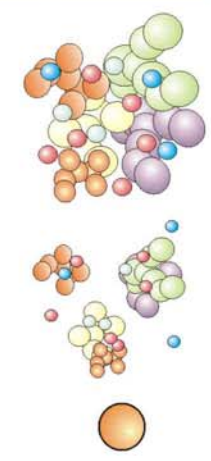
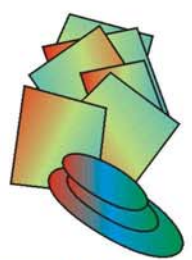
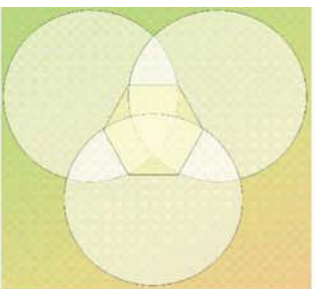


group BioGeomics

coevolution of biochemistry and natural organic matter (NOM)



molecular-level structural analysis of NOM



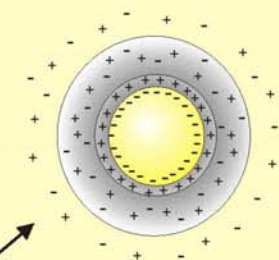
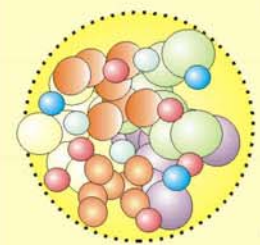
capillary electrophoresis

solution behavior, charge density, size and charge distributions, weak interactions

mass spectrometry

*molecular formula of all ions present
complex mixture analysis
Information concerning structure and reactivity*

the retention time / electrophoretic mobility carries itself (independent) structure-specific information

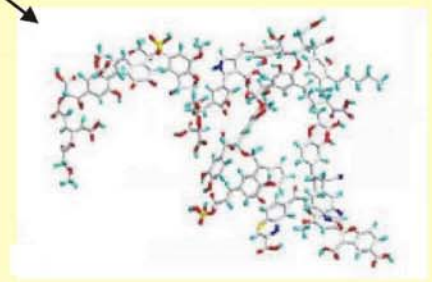


determination and assessment of significance of chemical structures, responsible and involved in a specific function of NOM

NMR spectroscopy

unsurpassed detailed information about short range molecular order

*multinuclear NMR
multidimensional NMR*



molecularly-resolved non-target analysis

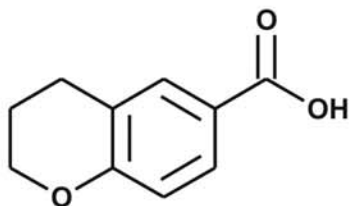
molecular diversity:

number of feasible isomeric molecules in one cubic meter

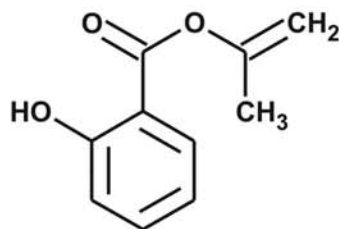
a typical molecule with nominal mass 178 Dalton $C_{10}H_{10}O_3$ (six DBE):

count of chemically relevant isomers: $1.1133 * 10^7$

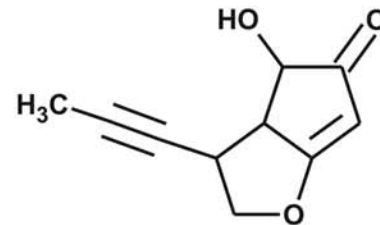
typical van der Waals volumes of $C_{10}H_{10}O_3$ isomers:



153.4 \AA^3



164.3 \AA^3

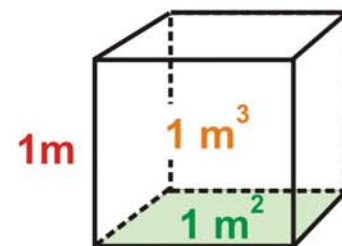


$C_{10}H_{10}O_3$

160.3 \AA^3

feasible count of $C_{10}H_{10}O_3$ molecules in one cubic meter

$$1 / 160 \text{ \AA}^3 \xrightarrow{V^{-1}} 6.3 * 10^{27} \text{ m}^{-3}$$

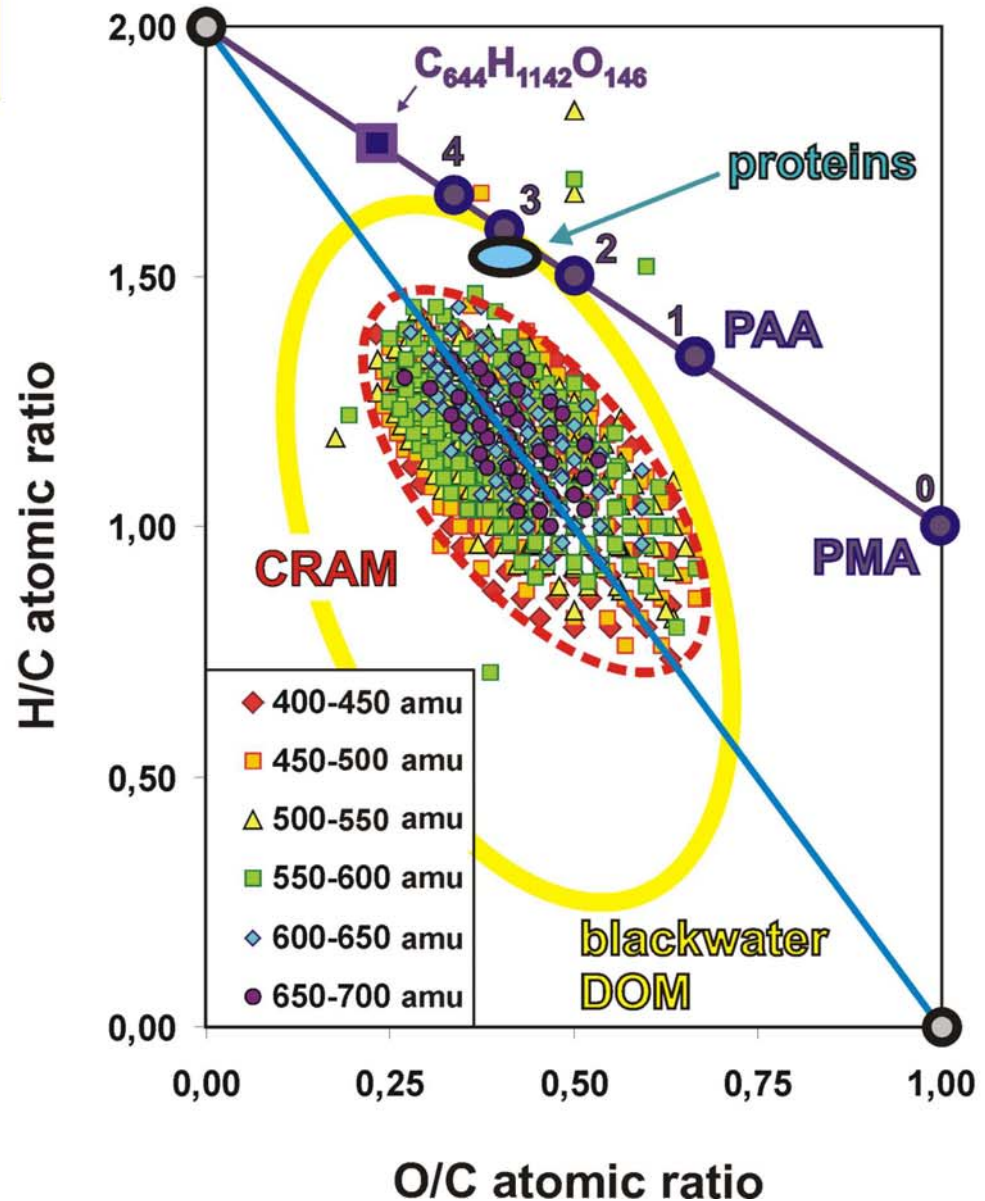


van Krevelen diagram

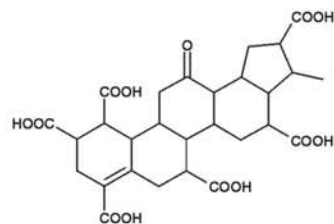
Visser, ES&T, 17 (1983) 412-417.

a powerful **visual**
representation
of complex mass
spectra, with far reaching
implications for the
structural analysis
of **NOM**

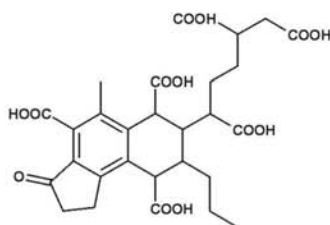
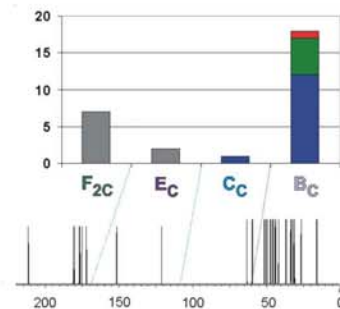
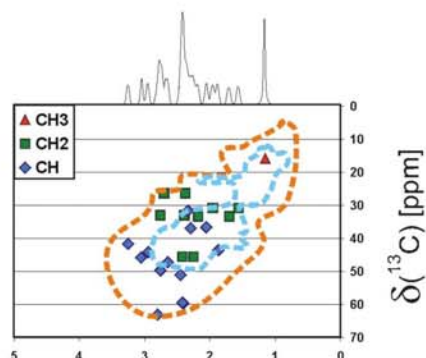
indicates unsaturation of **CRAM**
in excess of carboxylic groups



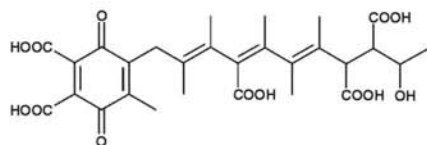
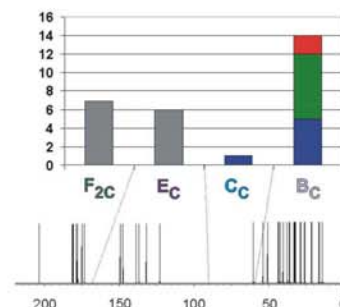
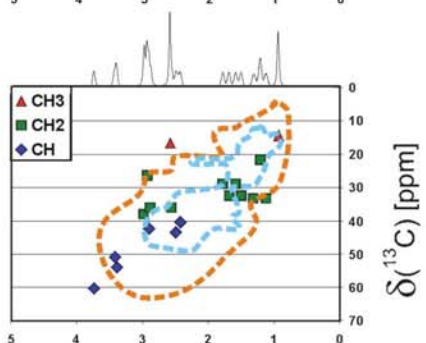
NMR properties serve to discriminate between (*classes* of) $C_{28}H_{32}O_{13}$ isomers (IUPAC mass: 576.546 Da)



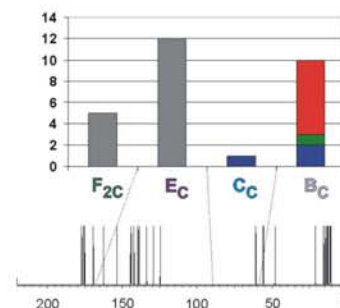
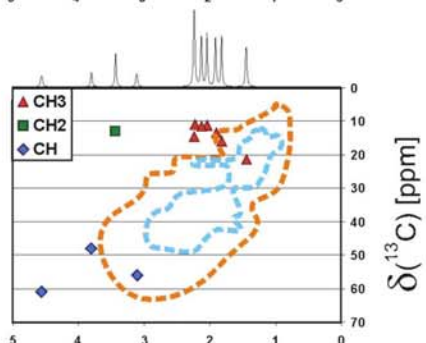
isomer I



isomer II



isomer III



$\delta(^1H)$ [ppm]

$\delta(^{13}C)$ [ppm]

potential significance of **CRAM**

major constituent of **marine UDOM**
and of the **global carbon cycle**

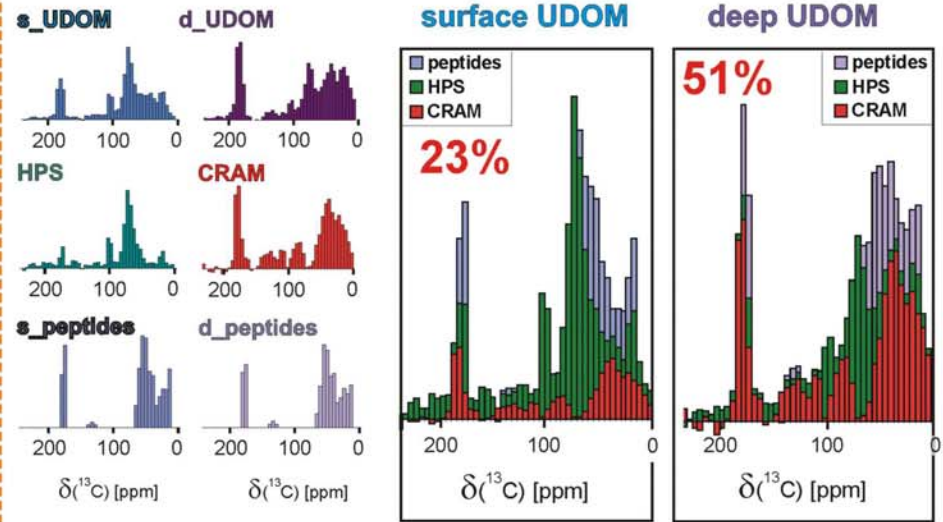
strong ligand for **metal binding**

may strongly assist in marine gel formation
controls bioavailability of

trace and nutrient **metals**

affects reactivity of **marine organic matter**

reverse mixing model from ^{13}C NMR data



CRAM is compositionally and structurally more **heterogeneous** than other **marine UDOM** constituents; this contributes to its resistance against biodegradation and to its **refractory** nature

*possible precursors of **CRAM***

terpenoids

(oxidized black carbon)

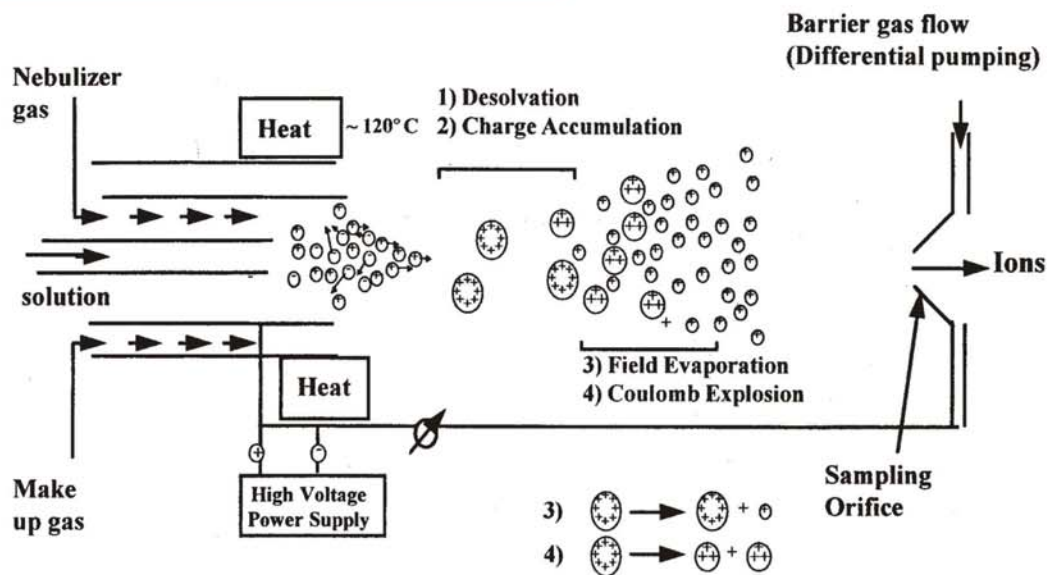
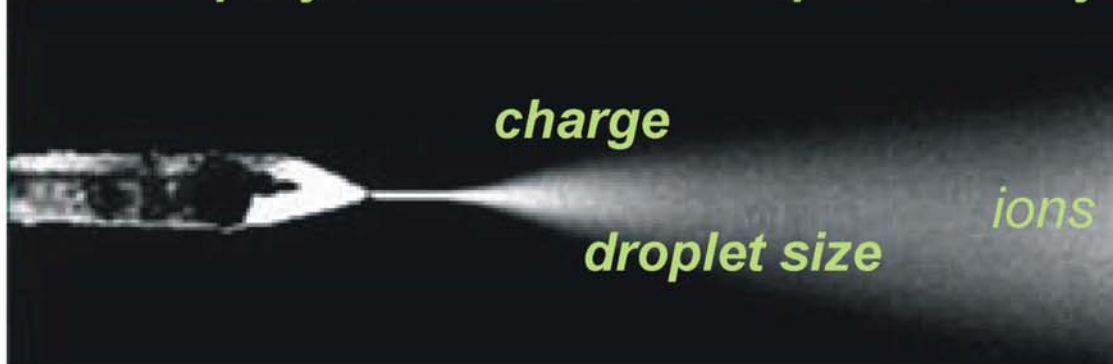
(((dimethylsulphonopropionate [DMSPP])))

the occurrence of **CRAM** in **freshwater** and **terrestrial** environments seems likely, considering the global distribution of biomolecules and the **similarity** of **biogeochemical processes** among environments

serious research on the **mass spectrometry of mixtures** is in it's very infancy

column adsorption
fractionation
electrochemistry
electrolysis
redox reactions

electrospray ionization mass spectrometry



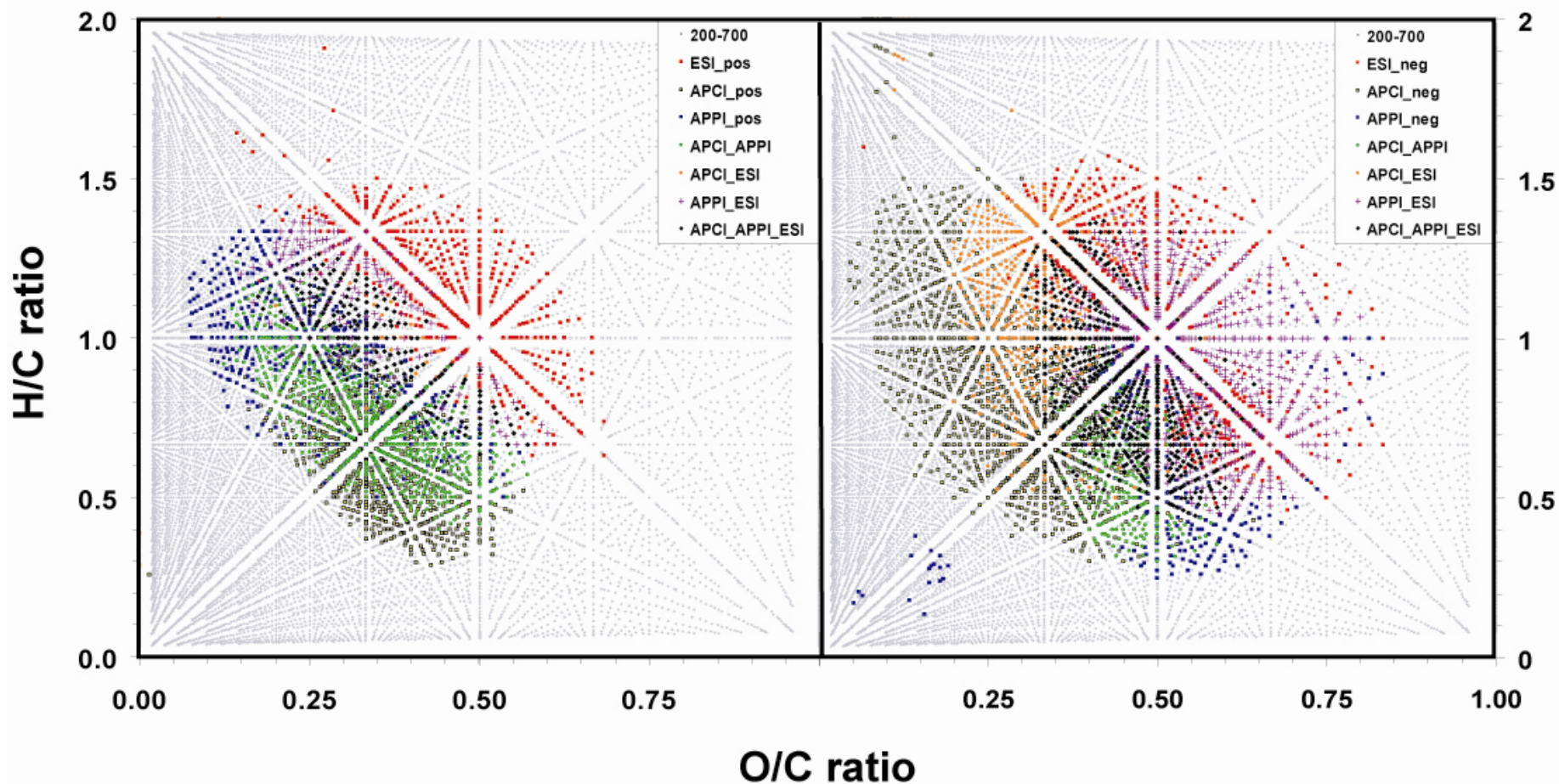
NOM
heterogeneity
polydispersity
ionization efficiency
ionization potential
metal ions
surface activity

FTICR mass spectrometry / complex systems

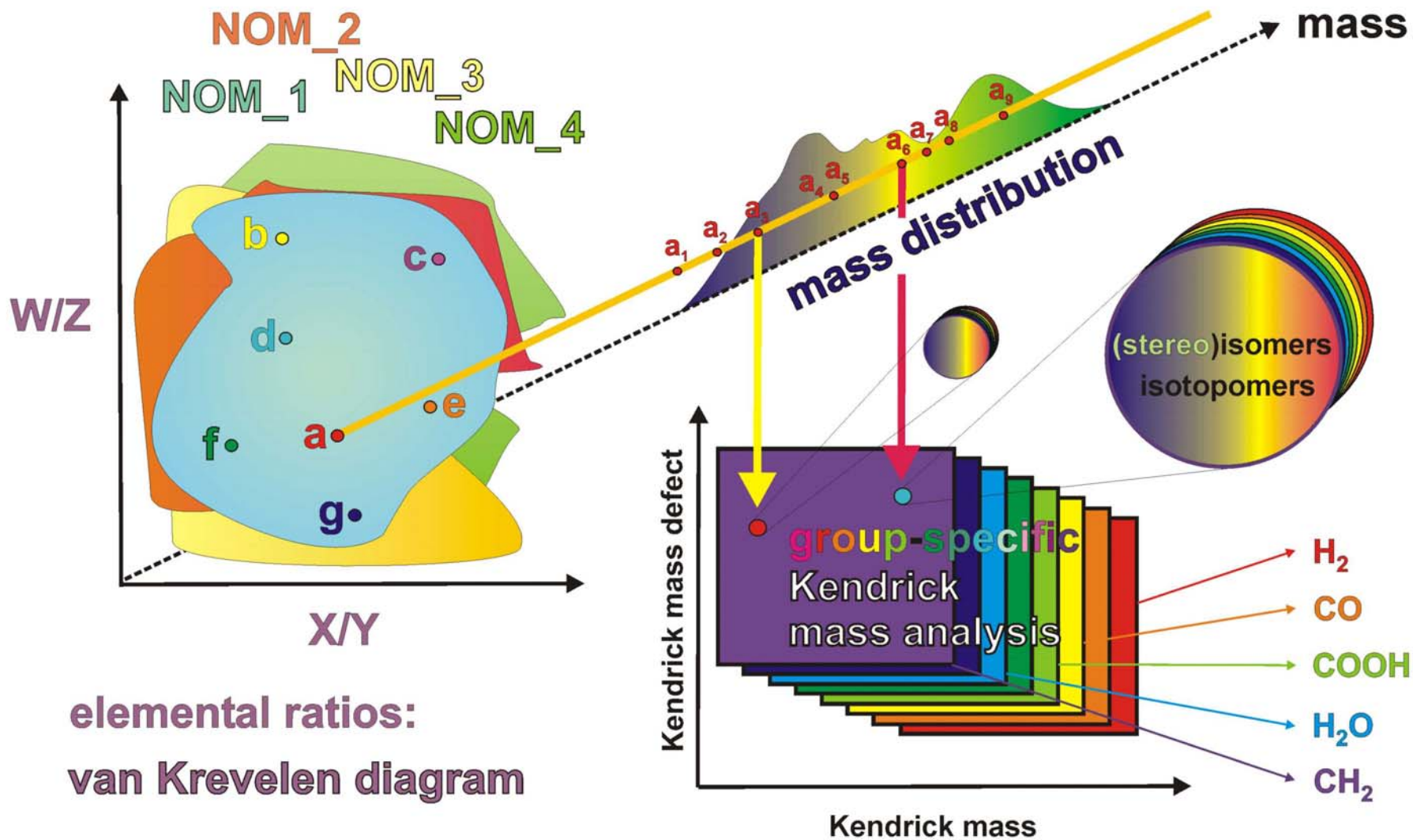
selectivity of ionization

positive

negative

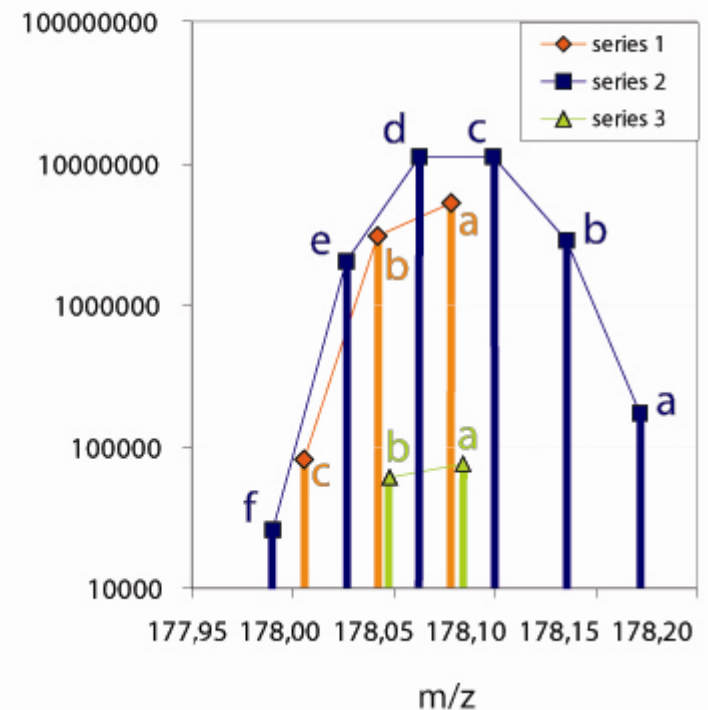
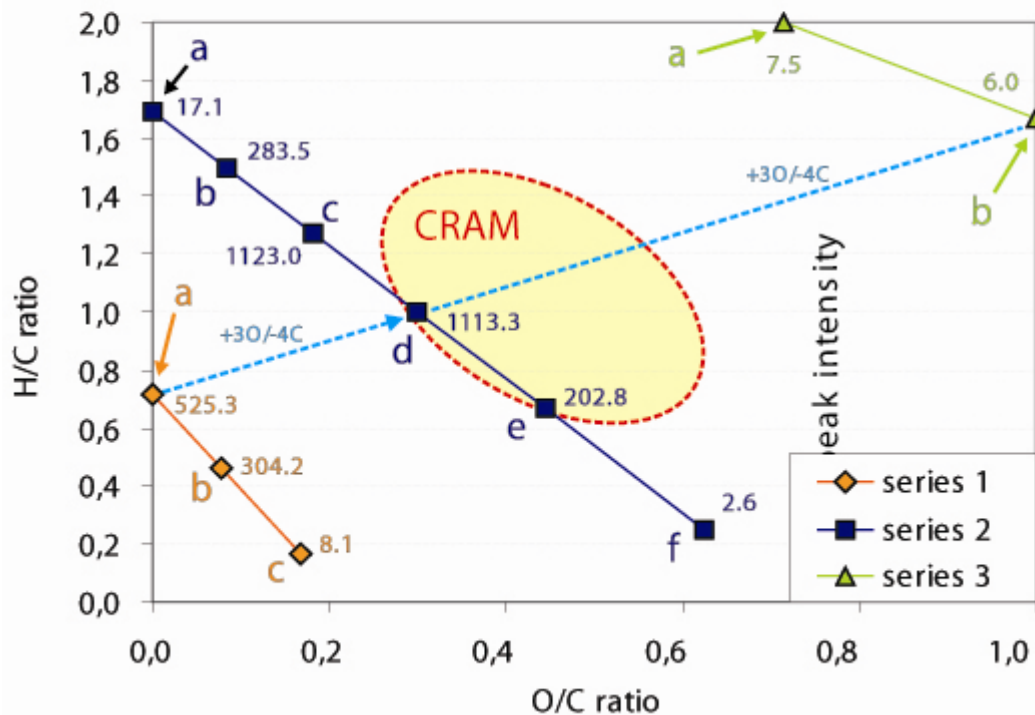


mass spectra confirm the immense complexity of NOM



FTICR mass spectrometry / complex systems

occurrence of mass spectral peaks relates with counts of chemically relevant structures

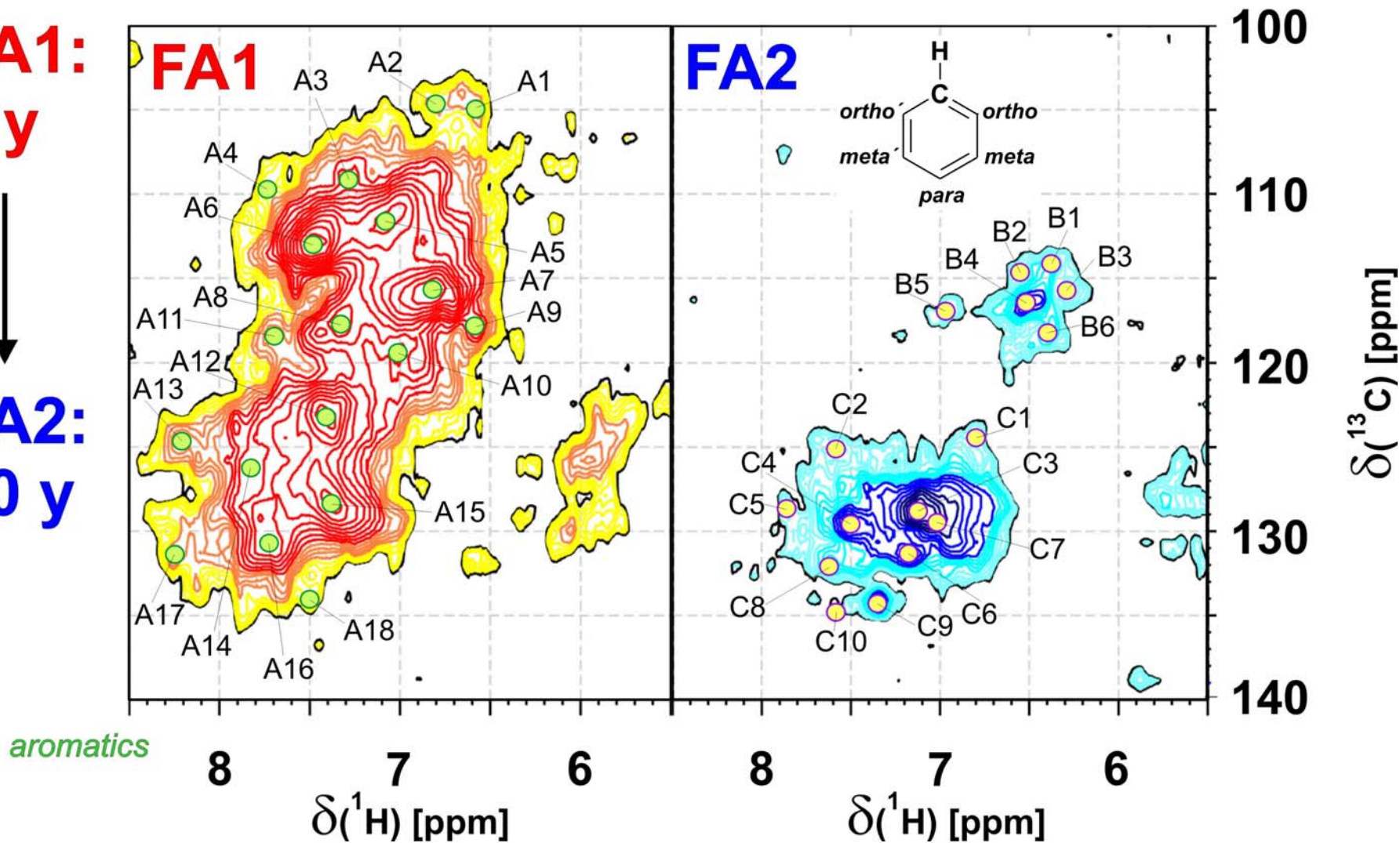


evolution of groundwater during 60 years

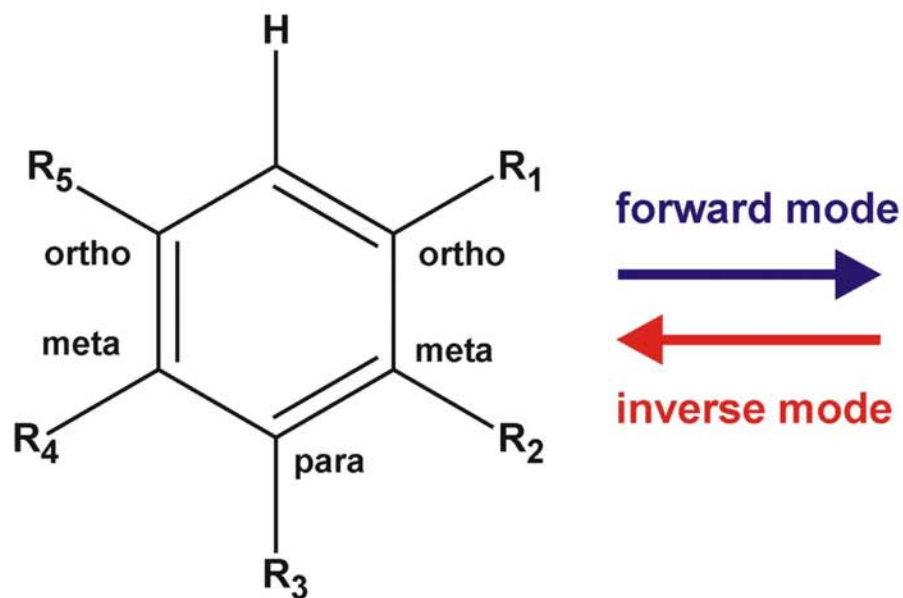
SPARIA analysis (Substitution Patterns in Aromatic Rings by Increment Analysis)

FA1:
0 y

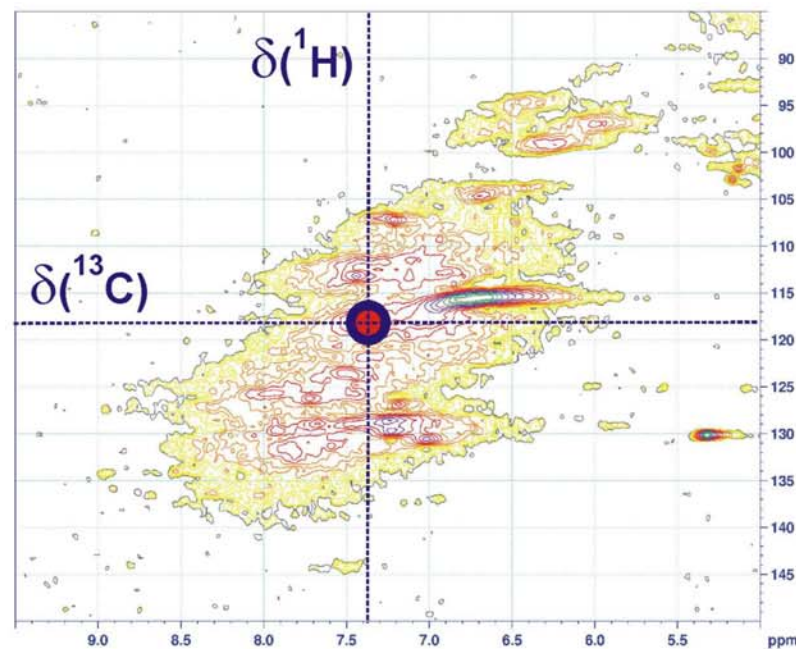
FA2:
60 y



Prediction of substitution patterns in aromatic rings by increment analysis (SPARIA)



¹H, ¹³C HSQC NMR of Suwannee River NOM



neutral	Ar-CO-X	Ar-O-X
-H	-COOCH ₃	-OCH ₃
-C ₂ H ₅	-COC ₂ H ₅	-OH
-CH=CH ₂	-COOH	

8 substituents in 5 positions =
32768 combinations